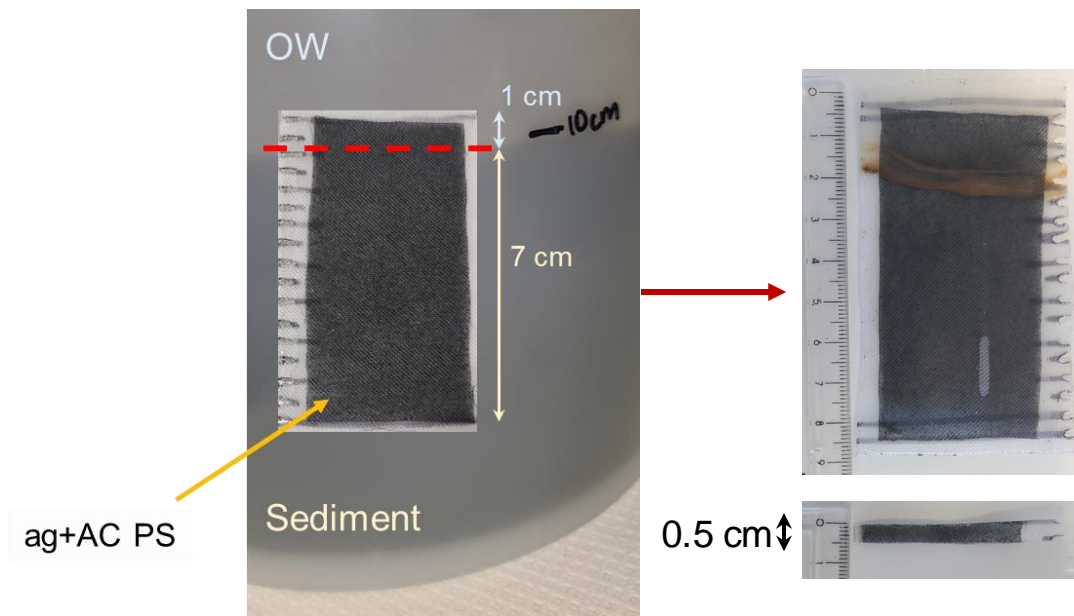


# Application of a Polymeric Equilibrium-Based Passive Sampler for Methylmercury to Measure a Sediment Porewater Depth Profile

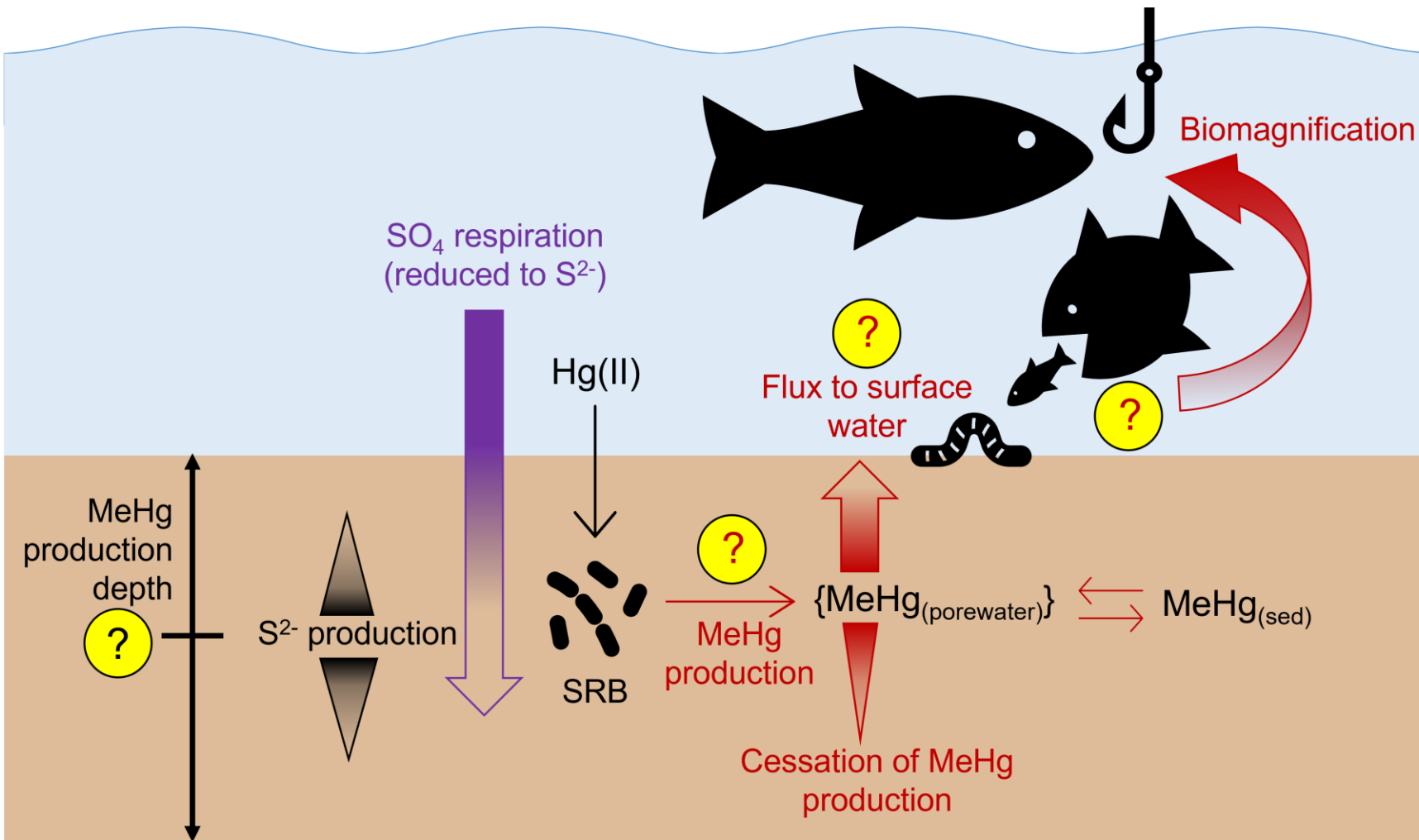
**Jada C. Damond<sup>[1]</sup>**, Cynthia C. Gilmour<sup>[2]</sup>, Upal Ghosh<sup>[1]</sup>

[1] Department of Chemical, Biochemical, and  
Environmental Engineering  
University of Maryland, Baltimore County  
Baltimore, MD

[2] Smithsonian Environmental Research Center  
Edgewater, MD



# Sampling difficulties limit understanding of Hg fate & transport *which is heavily influenced by the site's biogeochemistry*



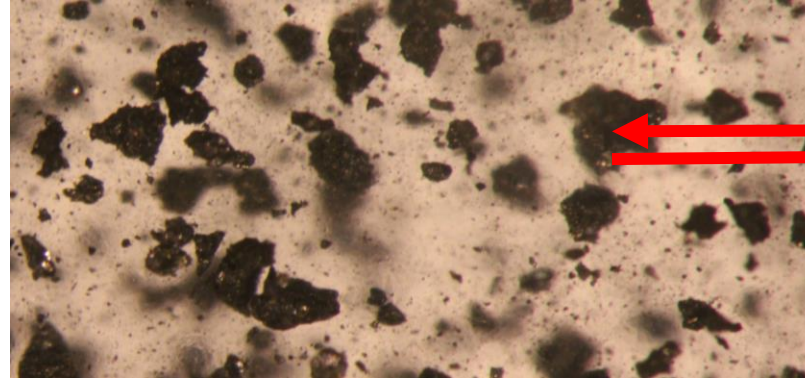
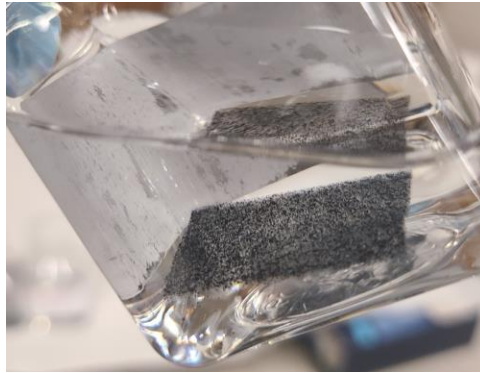
## Traditional Sampling Difficulties

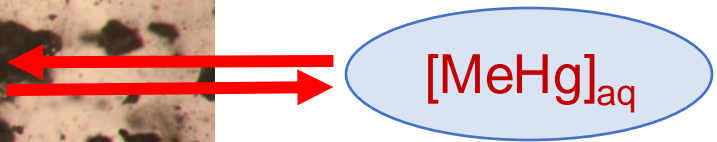
- **Low [MeHg]<sub>aq</sub>**: large sample needed for detection
- **Redox Sensitivity**: after collection, sediment core oxygenation influences Hg sediment:porewater dynamics and redox chemistry
- **Temporal Variability**: natural events can cause day-to-day variations, which may necessitate continuous sampling
- **Spatial Variability**: porewater MeHg reaches a peak somewhere down the porewater column

# Activated Carbon *Equilibrium* Passive Sampler (PS) developed to overcome challenges with traditional sampling methods

## ag+AC

Activated carbon suspended in agarose gel sheet (1 mm thick)

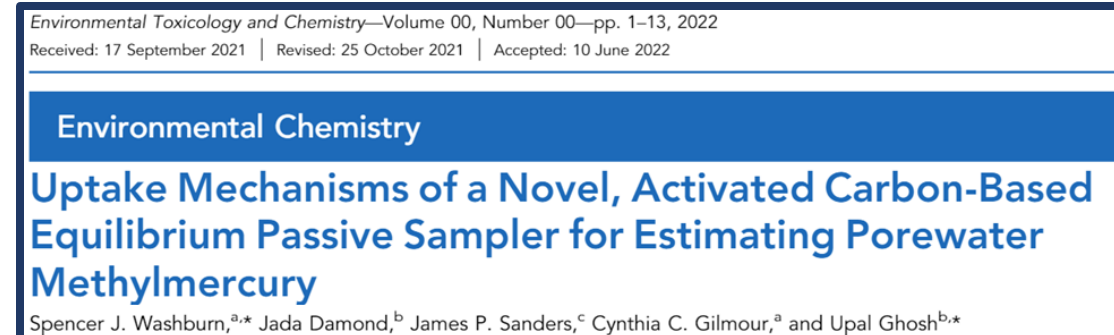
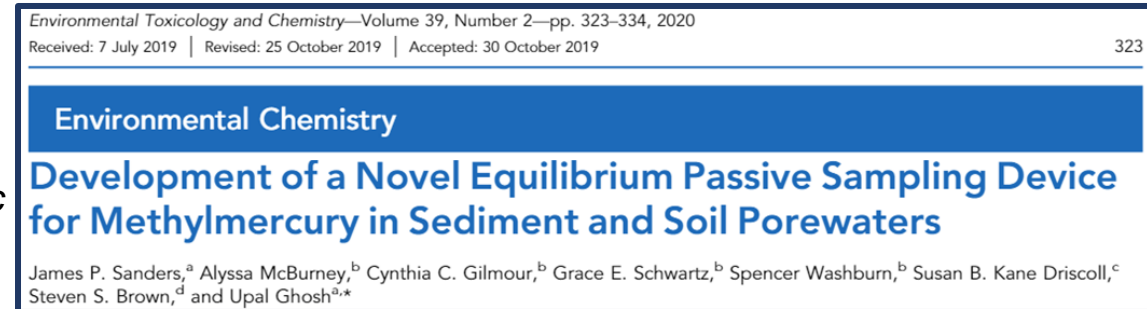



$$[\text{MeHg}]_{\text{aq}} = \frac{[\text{MeHg}]_{\text{PS}}}{K_{\text{PS}}}$$

## Demonstrations from Current Publications

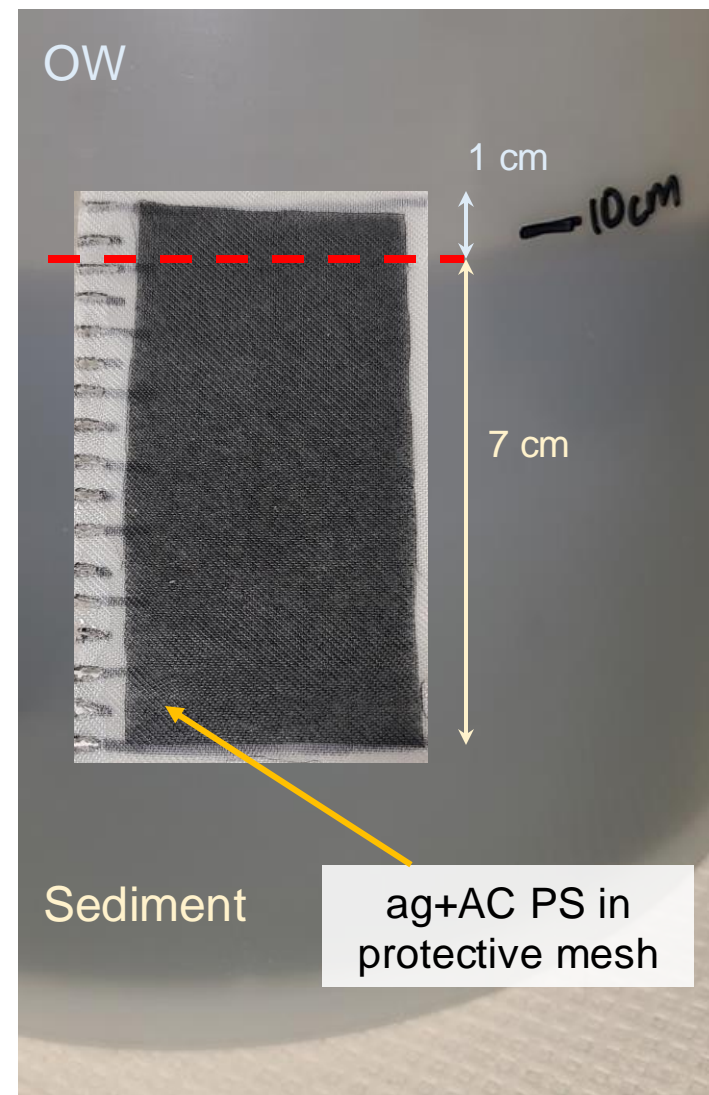
- **Improved Detection** ( $K_{\text{PS}} \cong 10^3$ )
- **Applicable in a variety of chemical environments**
  - *Tight  $K_{\text{PS}}$  range for MeHg complexes with several aquatic dissolved organic matter of varying character*
  - *minimal effects on sediment-PW dynamics*
- **Time-averaged measurements**
  - *PS-water reversible equilibrium reached in 14-28 days*
- **Measured a 2-point spatial  $[\text{MeHg}]_{\text{aq}}$  profile**
  - *using a single polymer placed across the sediment-water interface. Comparable w/in a factor of 4 to grab sample measurements*

**Next Step:** fine-scale porewater profiling





# Demonstration 1: measure a $[\text{MeHg}]_{\text{aq}}$ profile (0.5 cm resolution) with ag+AC PS and observe corresponding biogeochemistry



Deployed 8 x 4 cm ag+AC PS in freshwater sediment microcosms

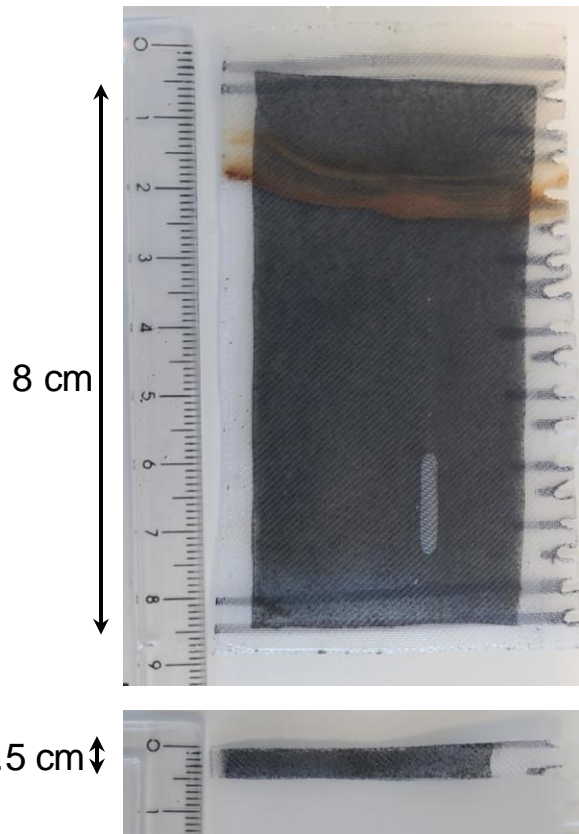
- Hg-laden sediment (BCSA) diluted with pristine freshwater sediment (SERC) & homogenized
- Overlying water (OW) collected from freshwater stream (SERC). Aerated throughout experiment
- 14-day incubation followed by 29-day PS deployment

## Initial Levels

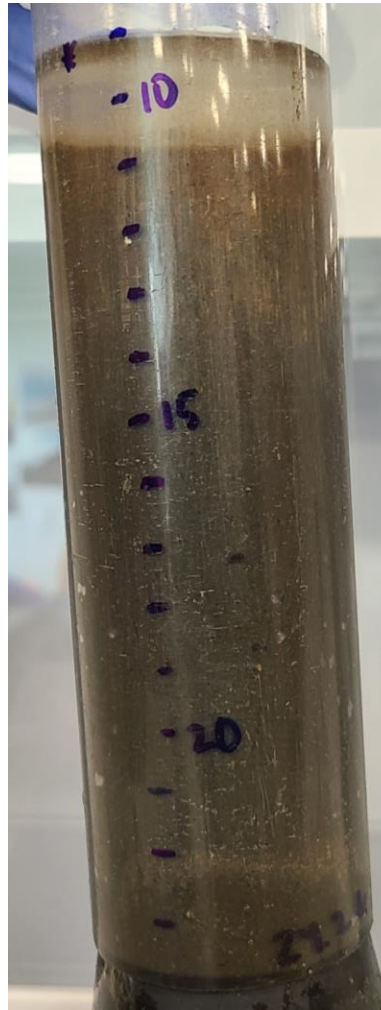
$[\text{THg}]_{\text{Sed}}$ (mg/kg)	$[\text{MeHg}]_{\text{Sed}}$ ( $\mu\text{g/kg}$ )	$[\text{MeHg}]_{\text{PW}}$ (ng/L)	$[\text{SO}_4^{2-}]_{\text{OW}}$ ( $\mu\text{M}$ )	$[\text{SO}_4^{2-}]_{\text{PW}}$ ( $\mu\text{M}$ )
4.4	2.3	0.56	154	2,854

# The ag+AC PS method was compared to collecting and sectioning sediment cores to provide a porewater profile

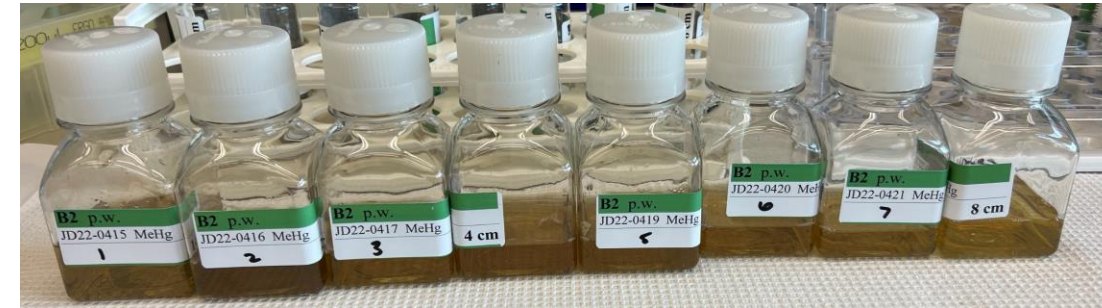
## Profiling with ag+AC PS



- VS -

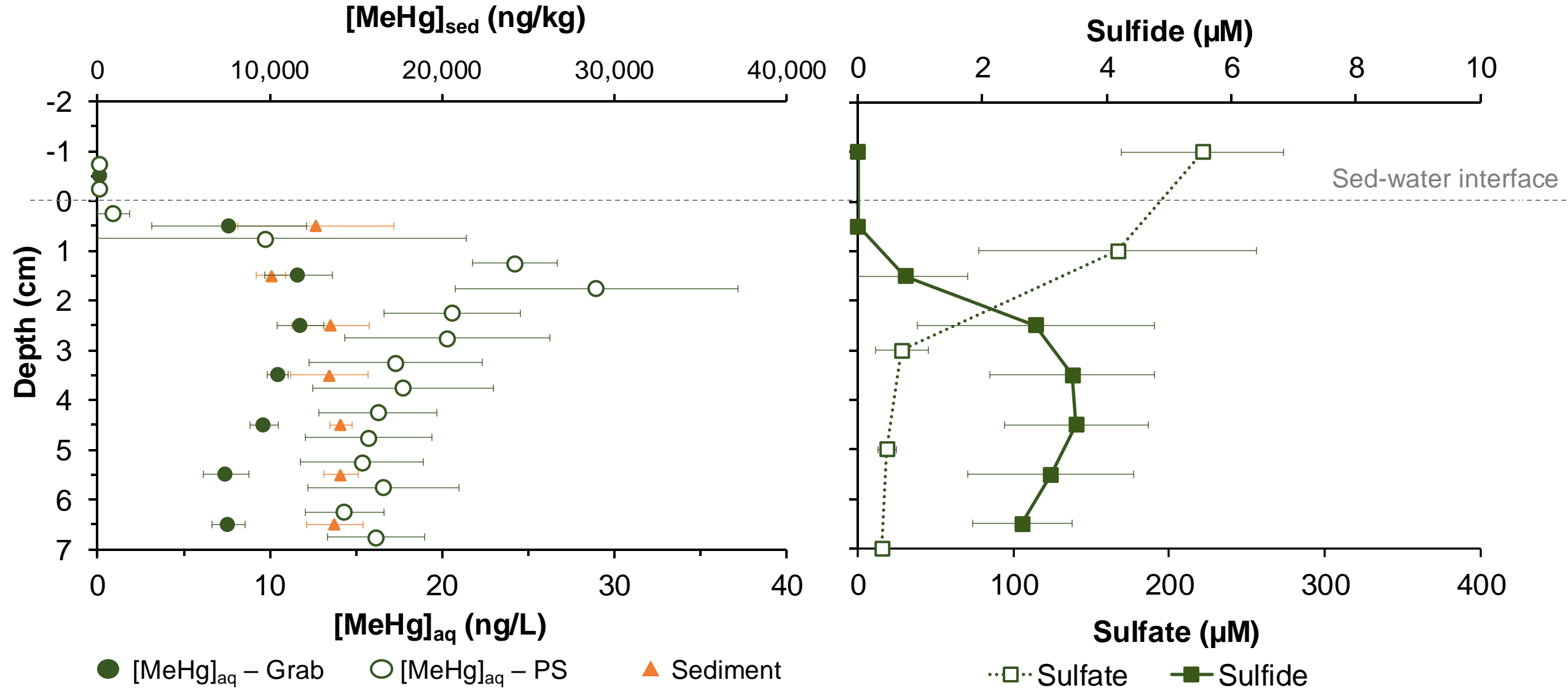


## Profiling with sediment cores & porewater **grab samples**



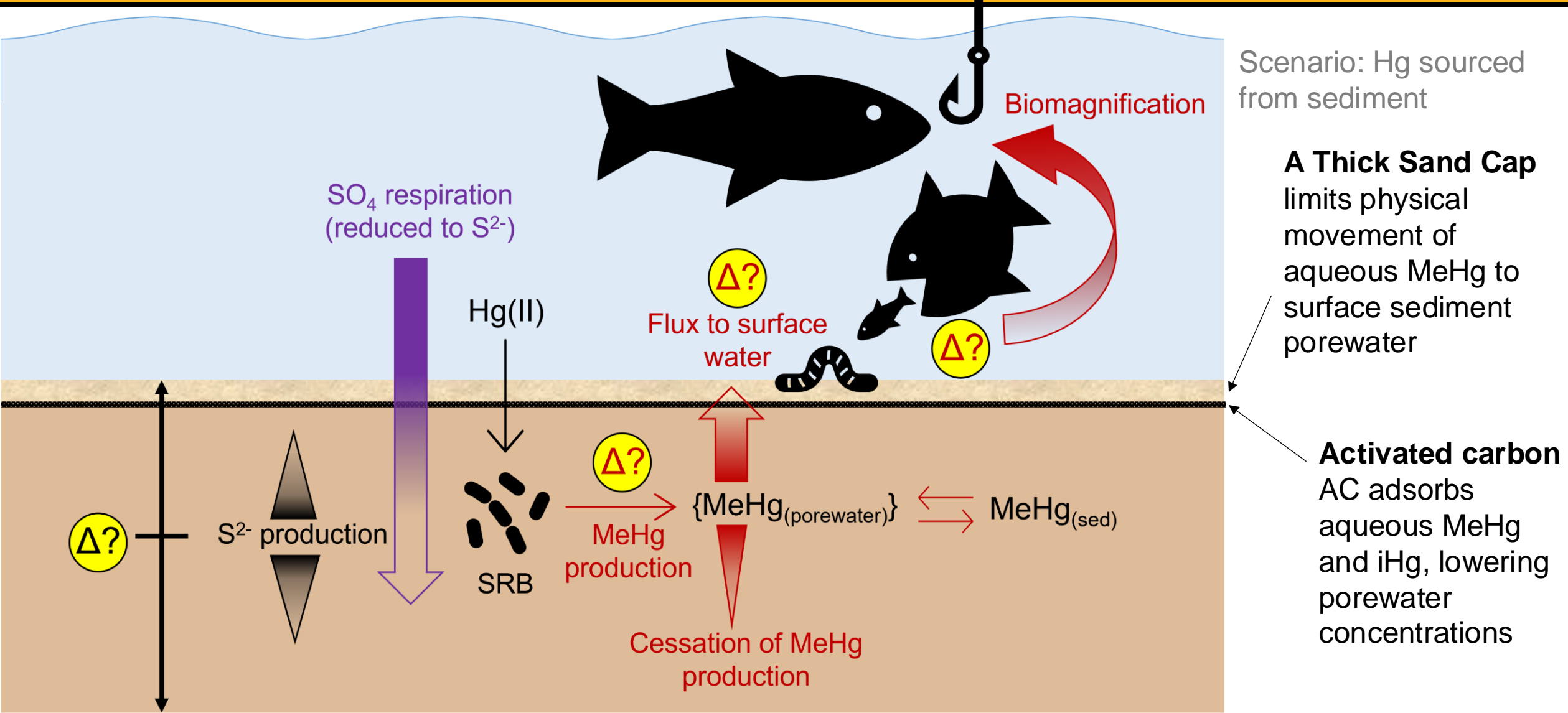
- *Several* cores collected from each microcosm
  - Needed to extract enough porewater for analyses
- Cores segmented (1 cm) in anaerobic glovebox
- Centrifugation extraction of porewater
  - Filtered to 0.45  $\mu\text{m}$

# Porewater profiles provide great insight into biogeochemical effects

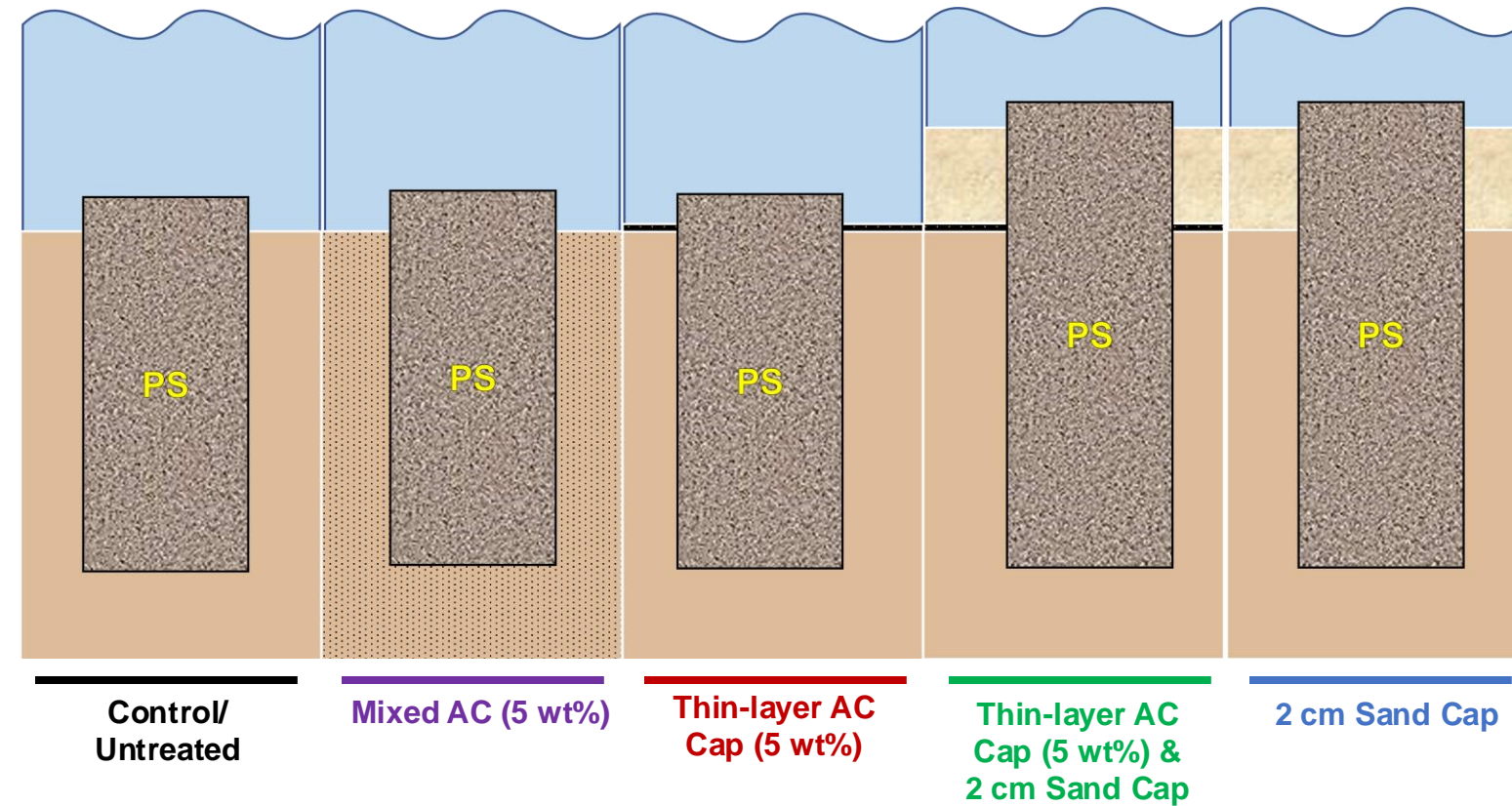




The goal for sediment treatment is to reduce MeHg bioavailability  
*i.e. by reducing surface porewater and surface water levels*



# Demonstration 2: use PS-Derived MeHg aqueous profiles to compare performance of several amendment applications and their effect on Hg fate & transport



Core from 2 cm Sand Cap

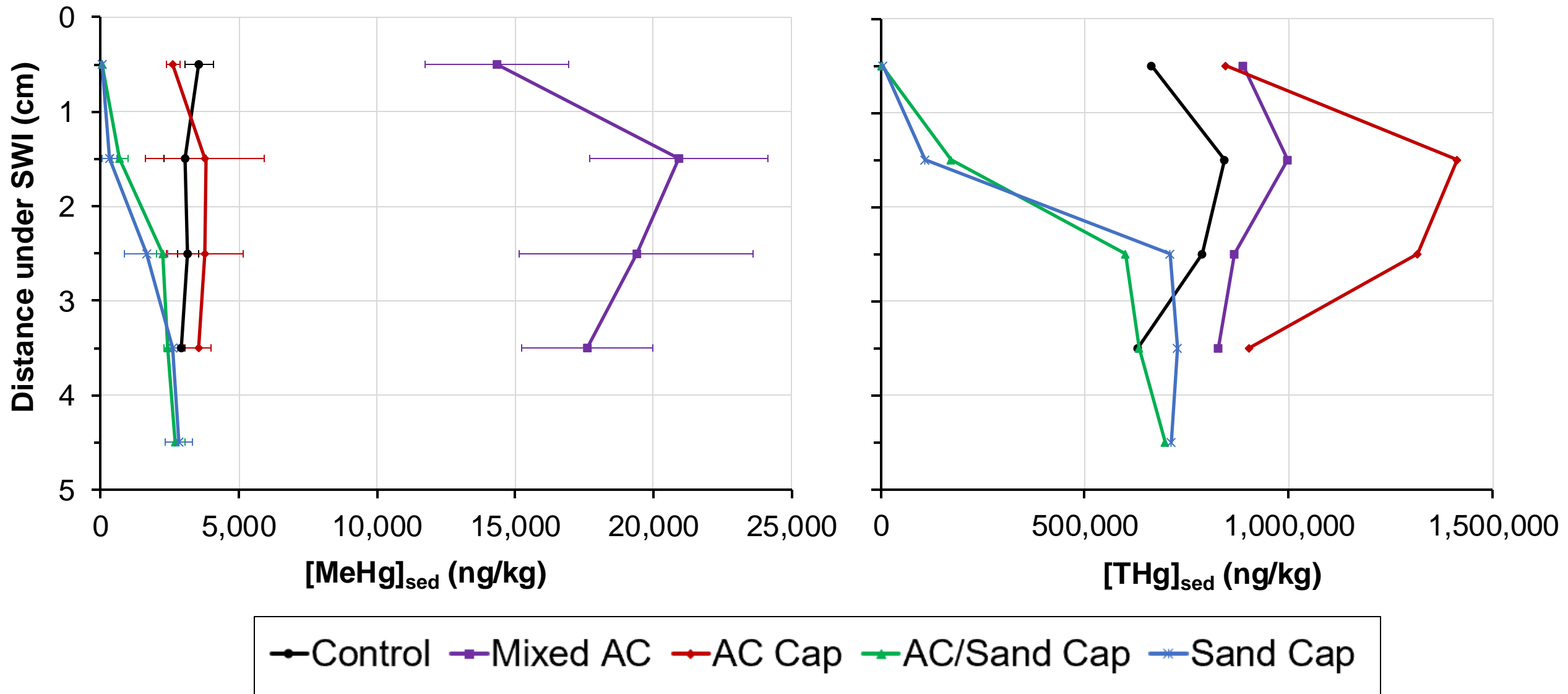


Core from Thin-layer AC Cap (5 wt%) & 2 cm Sand Cap

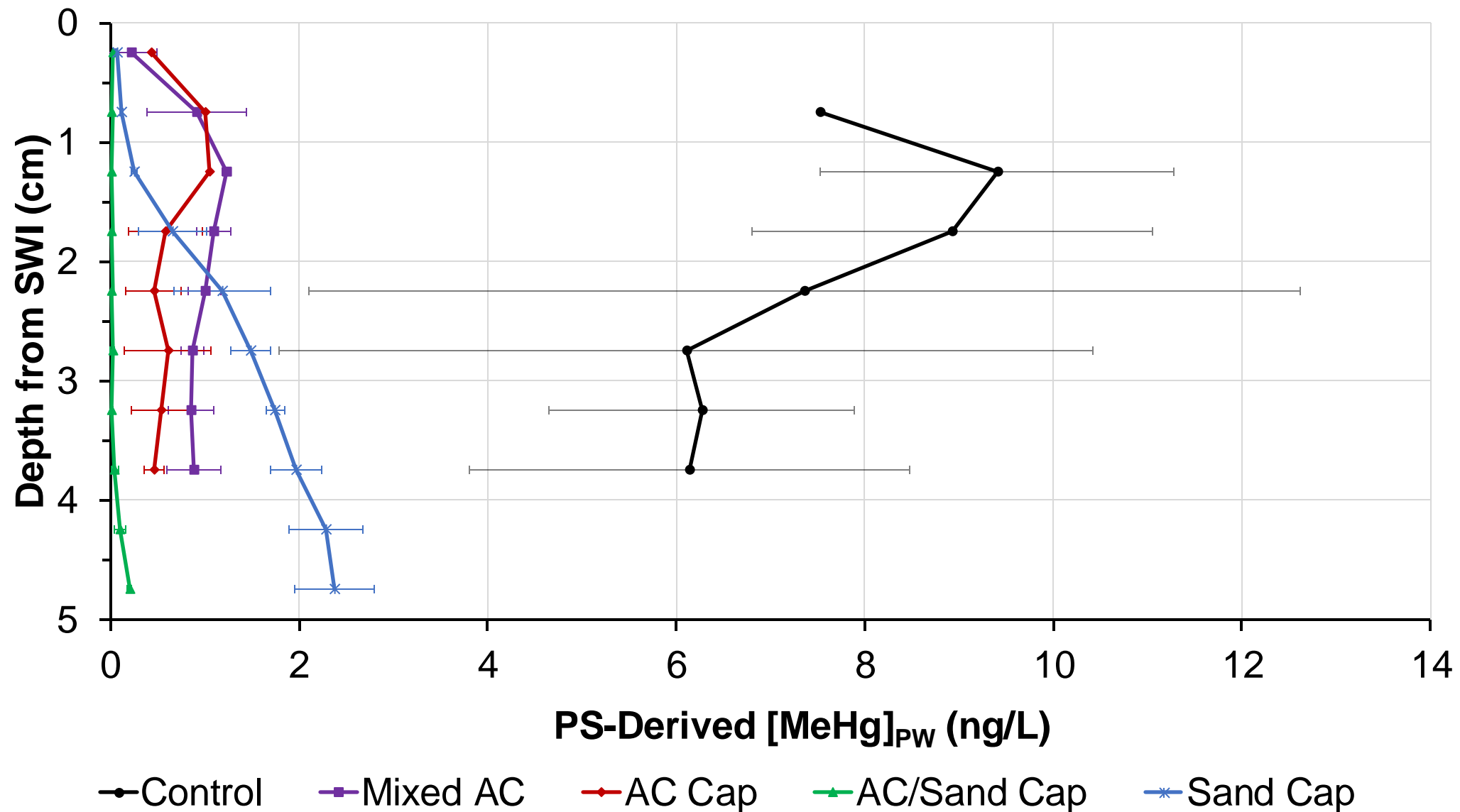
- Microcosms contain Hg-laden sediments (BCSA) diluted with pristine local freshwater sediments (SERC)
- Surface water collected from pristine local stream (SERC). Aerated throughout experiment
- 9 week incubation followed by 4 week PS deployment



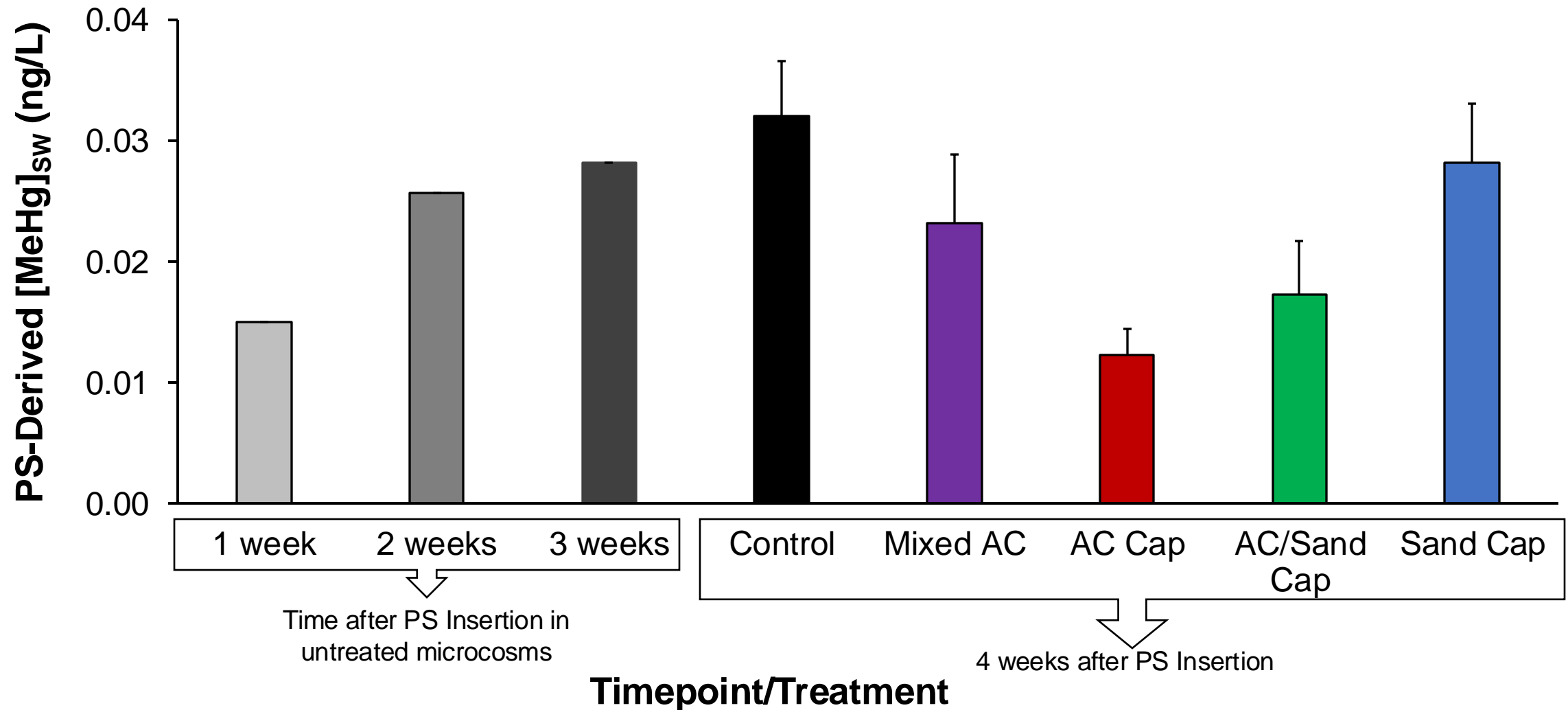
# Preliminary Results: Sediment MeHg and THg profiles



# Preliminary Results: PS-Derived porewater Profiles



# Preliminary Results: PS-Derived Surface Water Measurements





# Conclusions, Implications & Next Steps

- Demonstrated successful application of ag+AC PS for fine-scale (0.5 cm resolution)  $[\text{MeHg}]_{\text{aq}}$  profiling
- In demonstration 1, the porewater profile showed that peak  $[\text{MeHg}]_{\text{PW}}$  corresponded, spatially, to where sulfate reduction and sulfate production was observed.
  - *Suggestive of where the biological activity responsible for Hg methylation takes place, providing a target for treatment*
  - *Locating peak  $[\text{MeHg}]_{\text{PW}}$  can be indicative of exposure to sediment dwelling organisms*
- In demonstration 2, the depletion of  $[\text{MeHg}]_{\text{PW}}$  and  $[\text{MeHg}]_{\text{sed}}$  in treatments with a 2 cm sand cap indicated limited MeHg transport in surficial sediments
  - *Low porewater recovery in sand layers, porewater profiling with PS can be crucial for assessing cap performance and breakthrough*
- In demonstration 2, significant increase in  $[\text{MeHg}]_{\text{sed}}$  when 5 wt% AC was mixed into bulk sediments, opposed to cap with same AC mass, although  $[\text{MeHg}]_{\text{PW}}$  in the two treatments were comparable
  - *treatment application method does impact performance and Hg F&T.*

## Next Steps:

- additional chemical profiling in remediation study to study biogeochemical impact of treatment applications
- use data to build quantitative relationships to describe Hg F&T

# Thanks for Listening!

## Acknowledgements

Dr. Upal Ghosh  
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Azmat Naseem  
Jawairia Amjad



Smithsonian  
*Environmental Research Center*



*Let's Connect!*

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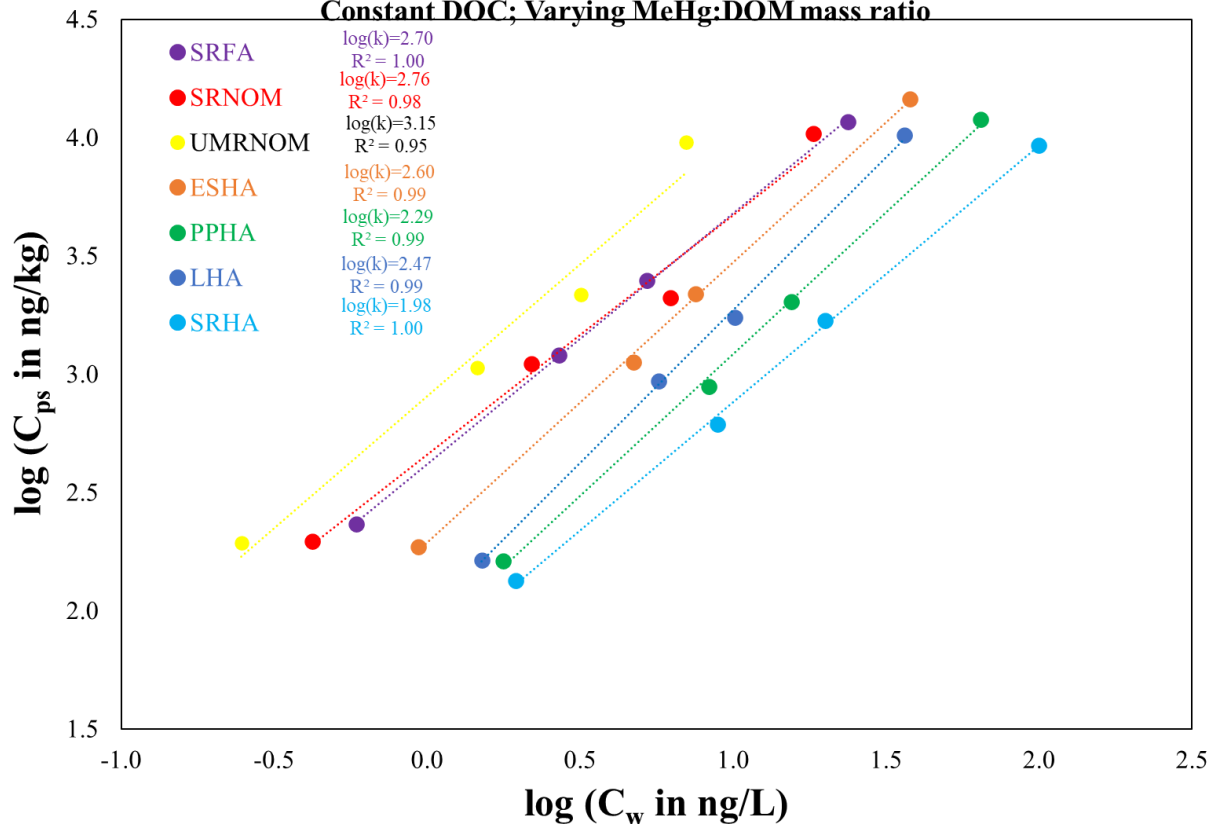
# Supplement Slides



# ag+AC PS Characterization & Development

## MeHgDOM partitioning to ag+AC (linear fit k)

Constant DOC; Varying MeHg:DOM mass ratio



- **Tight  $K_{PS}$  range** for aquatic, unfractionated HMW and LMW DOM species
- **Detectable Measurements** ( $K_{PS} \cong 10^3$ )  
Using average  $K_{PS}$  from DOM isotherms
- ag+AC **equilibrium reached after 14 days**

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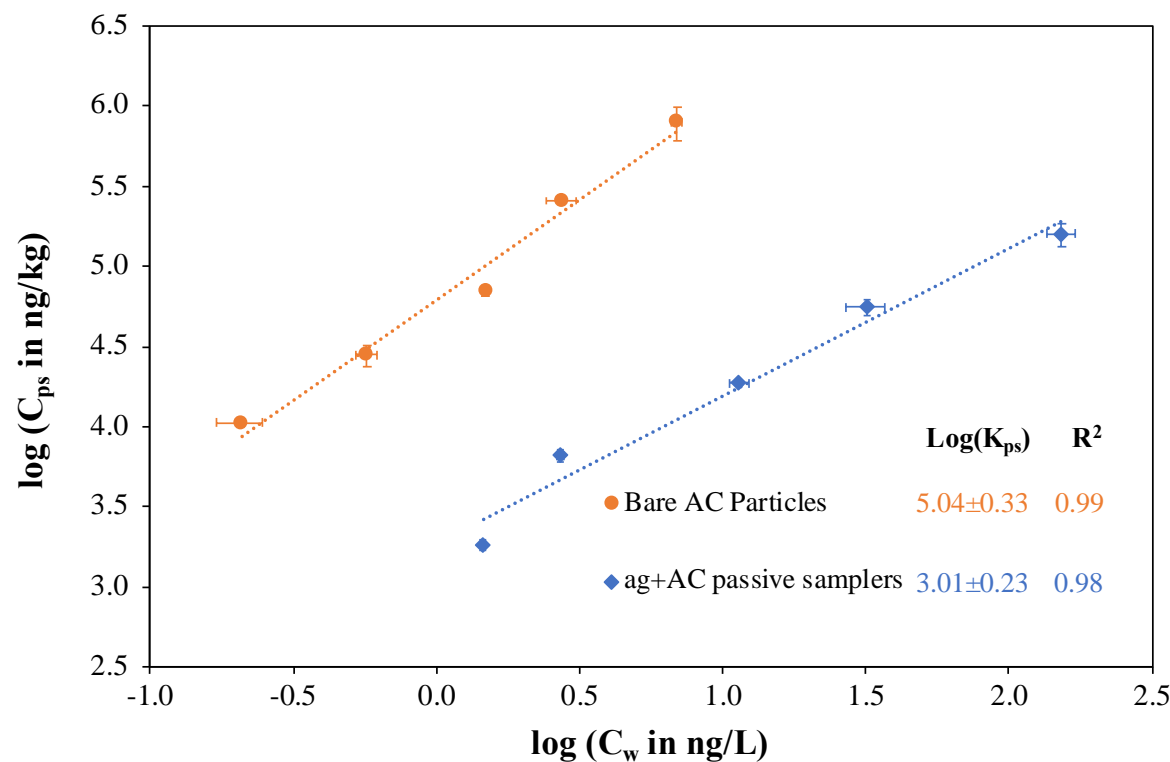
Received: 17 September 2021 | Revised: 25 October 2021 | Accepted: 10 June 2022

## Environmental Chemistry

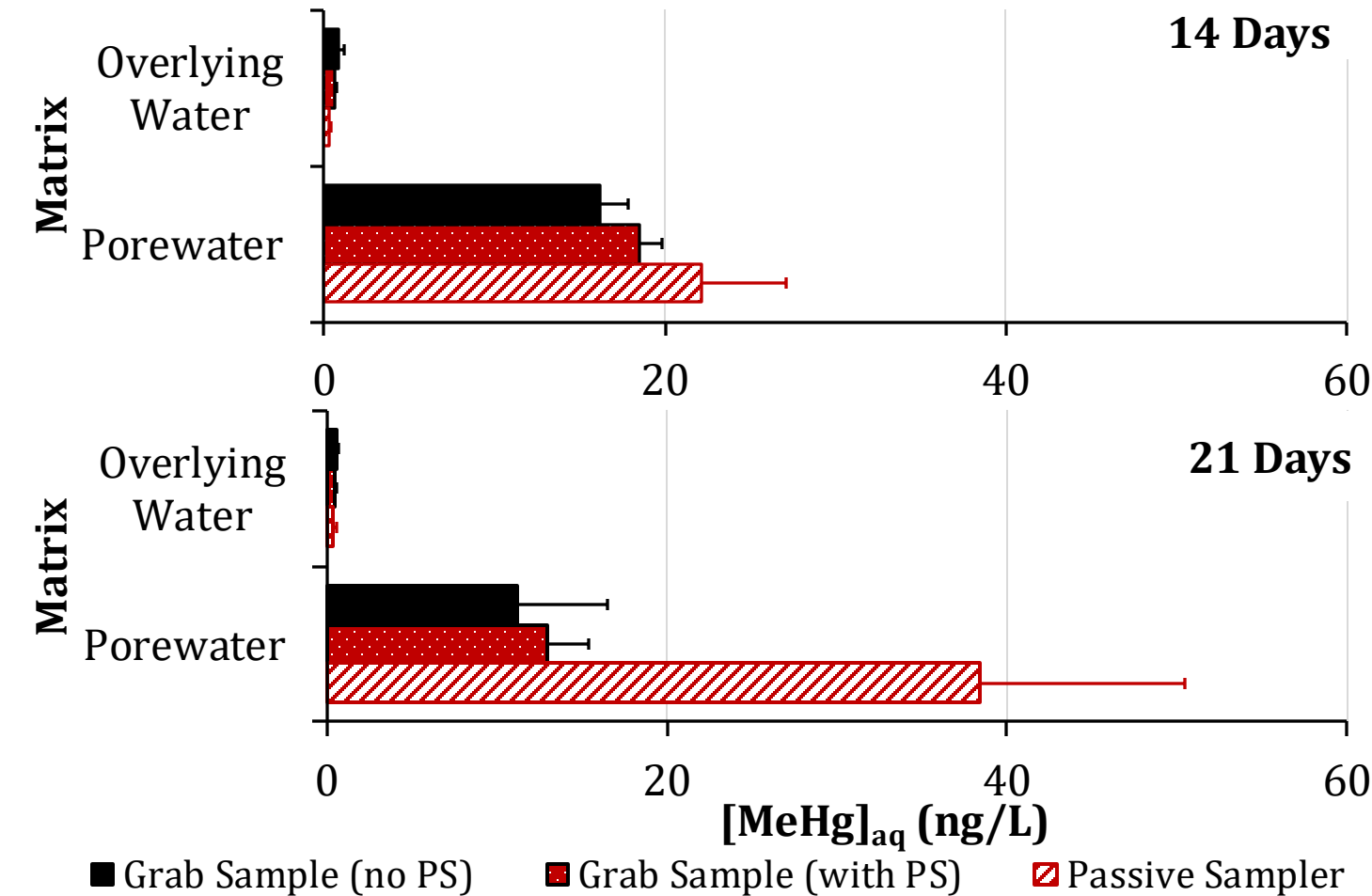
## Uptake Mechanisms of a Novel, Activated Carbon-Based Equilibrium Passive Sampler for Estimating Porewater Methylmercury

Spencer J. Washburn,<sup>a,\*</sup> Jada Damond,<sup>b</sup> James P. Sanders,<sup>c</sup> Cynthia C. Gilmour,<sup>a</sup> and Upal Ghosh<sup>b,\*</sup>

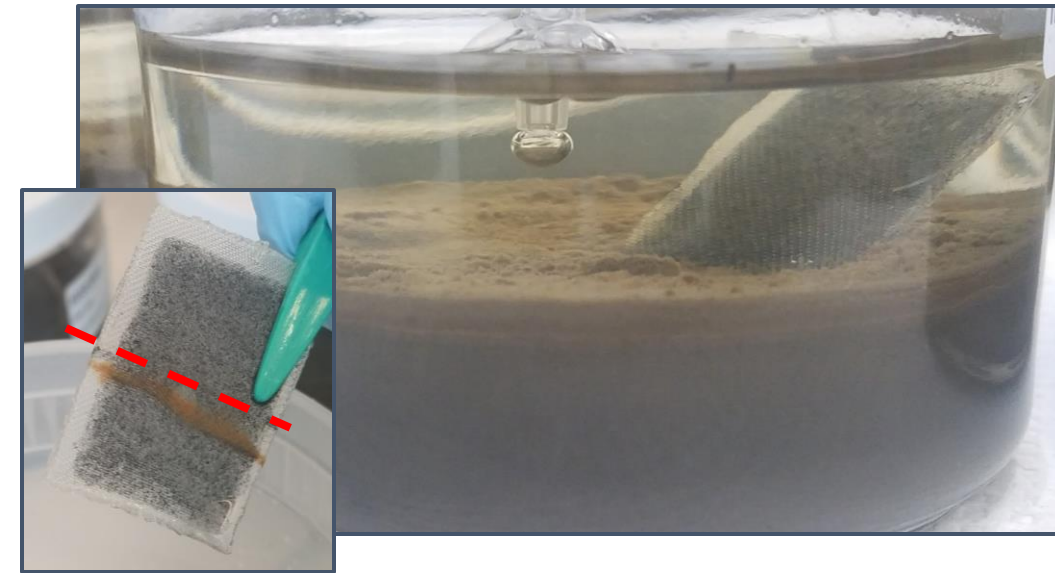
## MeHgCys Sorption Isotherms



# ag+AC PS successfully measured time-integrated $[\text{MeHg}]_{\text{aq}}$ across sed-water interface



$$[\text{MeHg}]_{\text{aq}} = \frac{[\text{MeHg}]_{\text{PS}}}{K_{\text{PS}} (= 10^3)}$$



- PS-Derived measurements **within a factor of 4** of grab sample measurements
- PS had minimal effects on sediment and porewater conc.