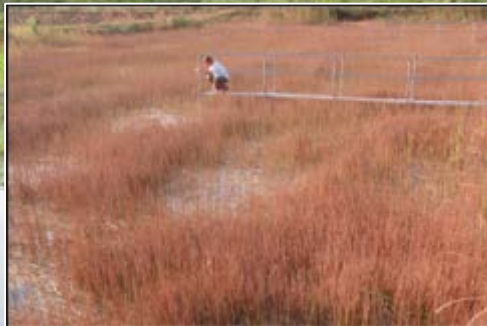


# Passive Treatment Systems for the Removal of Selenium: Selenium Removal Mechanisms with Biochemical Reactor Substrate



R.C. Thomas, J.J. Tudini, J.S. Bays, M.A. Girts, K.B. Jenkins,  
L.C. Roop, and T. Cook

**CH2MHILL®**

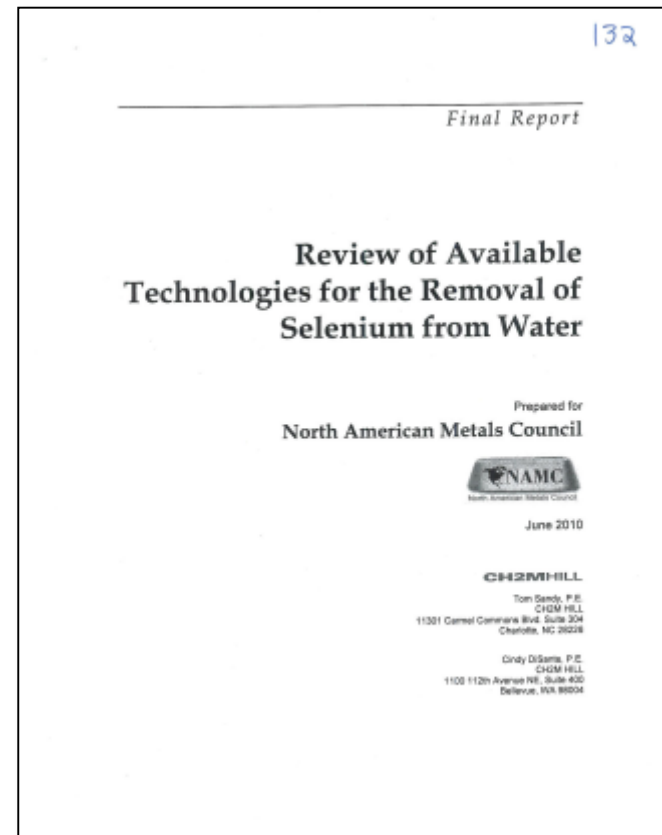
*June 3 2013*

# Overview of Presentation

## Outline

- Barrel-Study Overview
- Aqueous Results
- Volumetric/Aerial Loading Rates
- Substrate Se Removal
- Volatilization
- Substrate Se Speciation
- Modeling Calibration/Removal Rate Development
- Secondary Parameters
- Conclusions

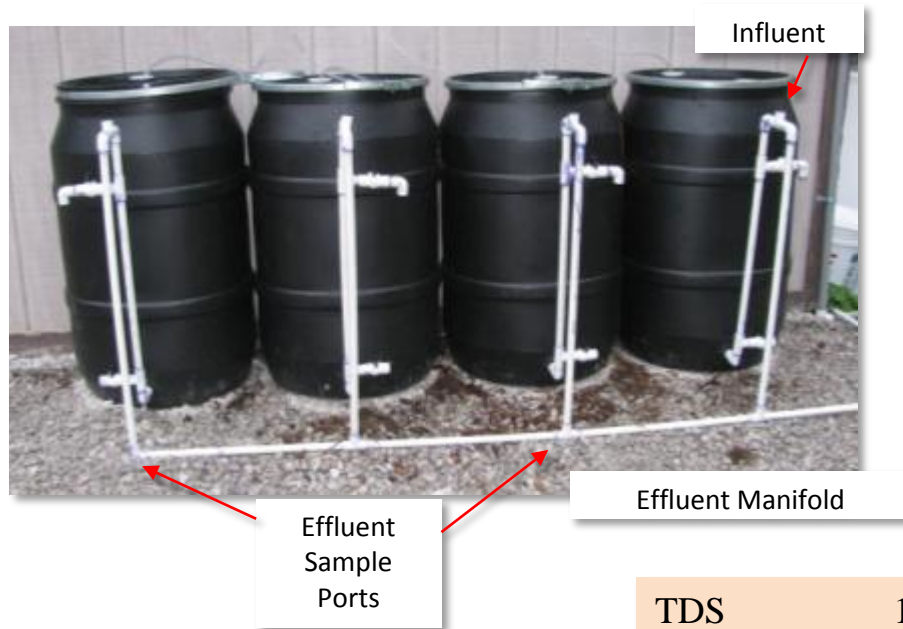
## NAMC Selenium Report 2010



<http://www.namc.org/docs/00062756.PDF>

# Case History (2011-2012) Two Concurrent BCR Pilot Studies for Coal Mine Drainage

**Outlet A: 389 days  
Pumped Inflow**



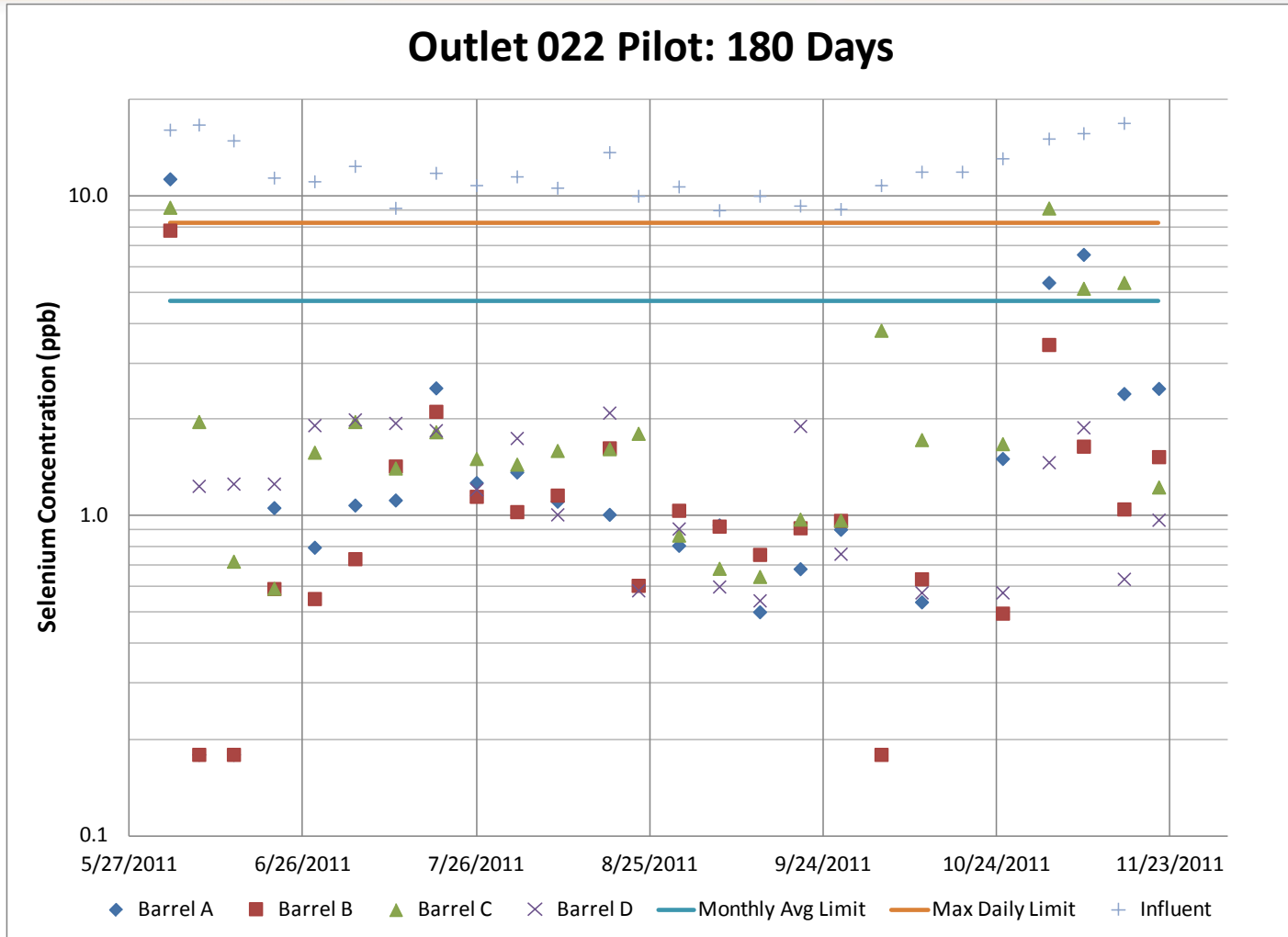
**Outlet B: 180 days  
Gravity Inflow**



TDS	1	g/L
Se in	6-17	µg/L
NO <sub>3</sub> -N	0.47	mg/L
TP	0.071	mg/L

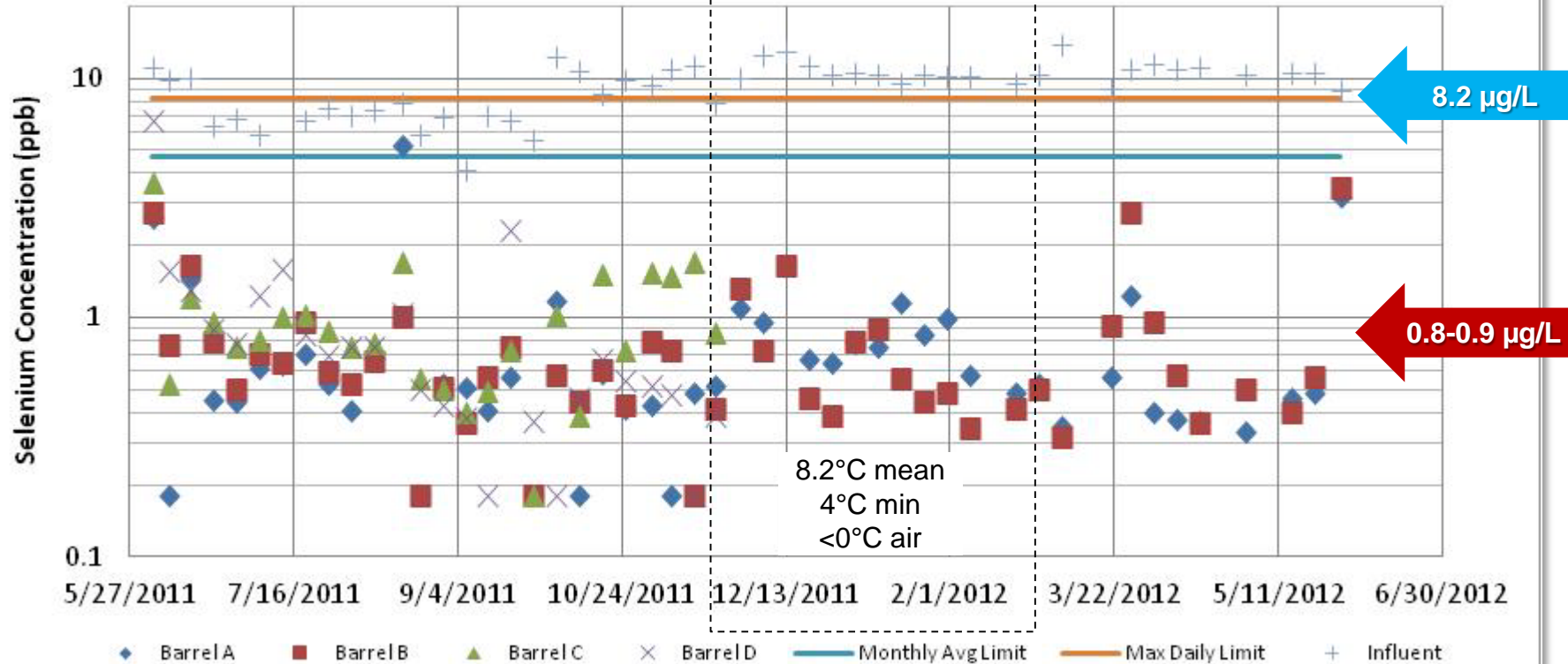
Source:  
R. Thomas (2011)

# Outlet 022: Variable Flow and Limited Flow Control Resulting in Variable Se Reduction

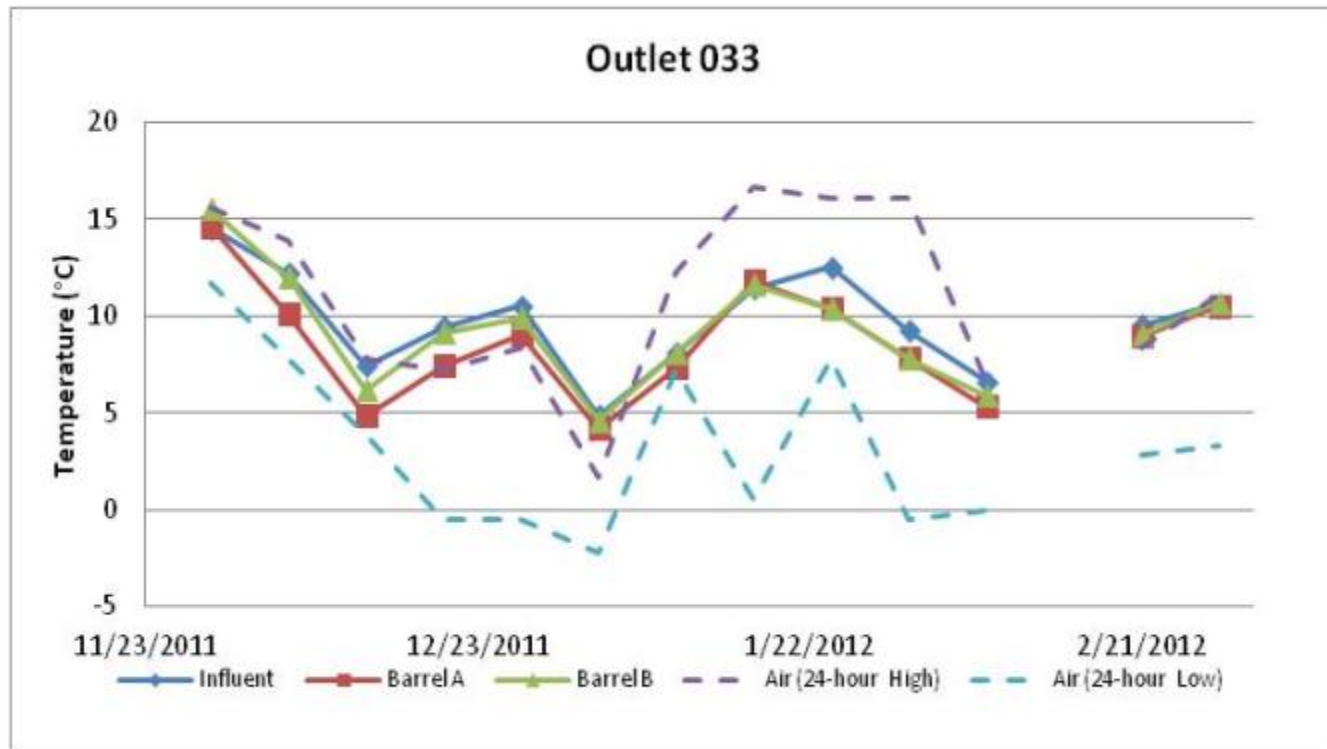


# Outlet 033: Consistent Se Reduction Through Winter

## Outlet 033 Pilot: 389 days



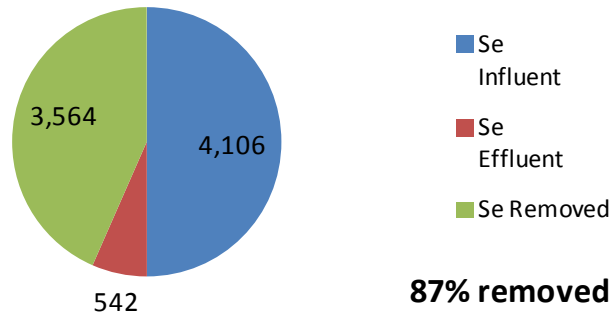
# Focus on Winter Water Minimum Temperature 4°C



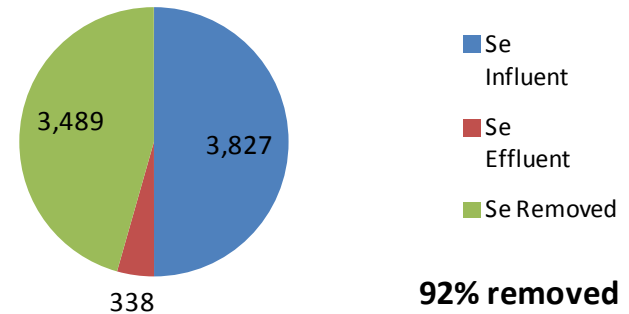
Avg Water Temp 8.2°C  
Avg Min Air Temp 4.2°C

# Average Se Mass Removed Per Day – 022

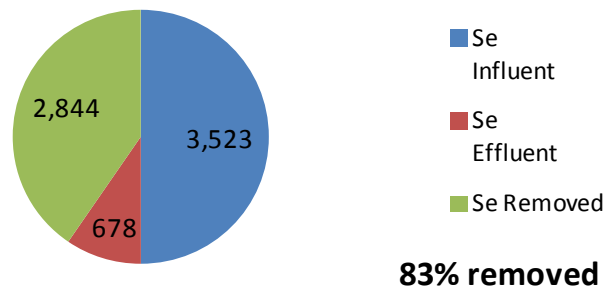
**Outlet 022 - Barrel A,**  
Avg. Se Mass ( $\mu\text{g}/\text{day}$ )



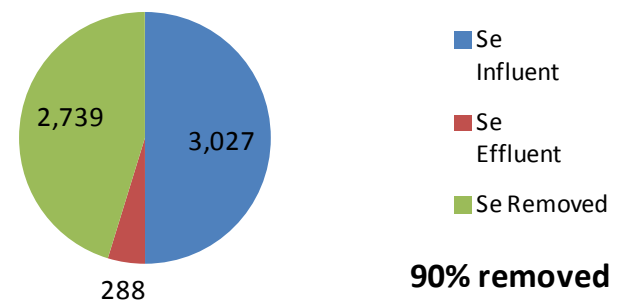
**Outlet 022 - Barrel B**  
Avg. Se Mass ( $\mu\text{g}/\text{day}$ )



**Outlet 022 - Barrel C**  
Avg. Se Mass ( $\mu\text{g}/\text{L}$ )

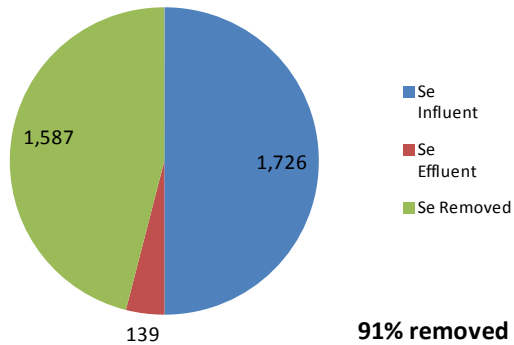


**Outlet 022 - Barrel D**  
Avg. Se Mass ( $\mu\text{g}/\text{L}$ )

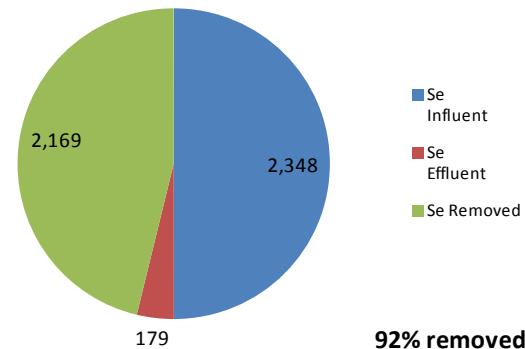


# Average Se Mass Removed Per Day – 033

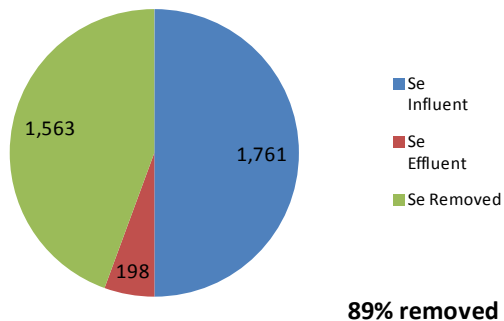
**Outlet 033 - Barrel A,**  
Avg. Se Mass ( $\mu\text{g}/\text{day}$ )



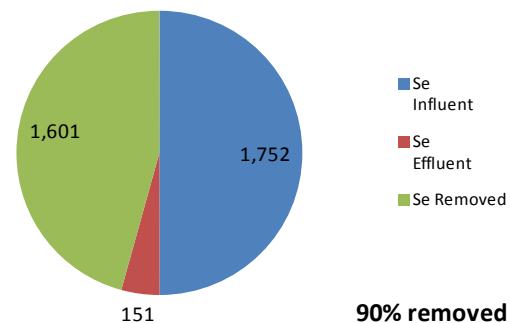
**Outlet 033 - Barrel B**  
Avg. Se Mass ( $\mu\text{g}/\text{day}$ )



**Outlet 033 - Barrel C**  
Avg. Se Mass ( $\mu\text{g}/\text{L}$ )

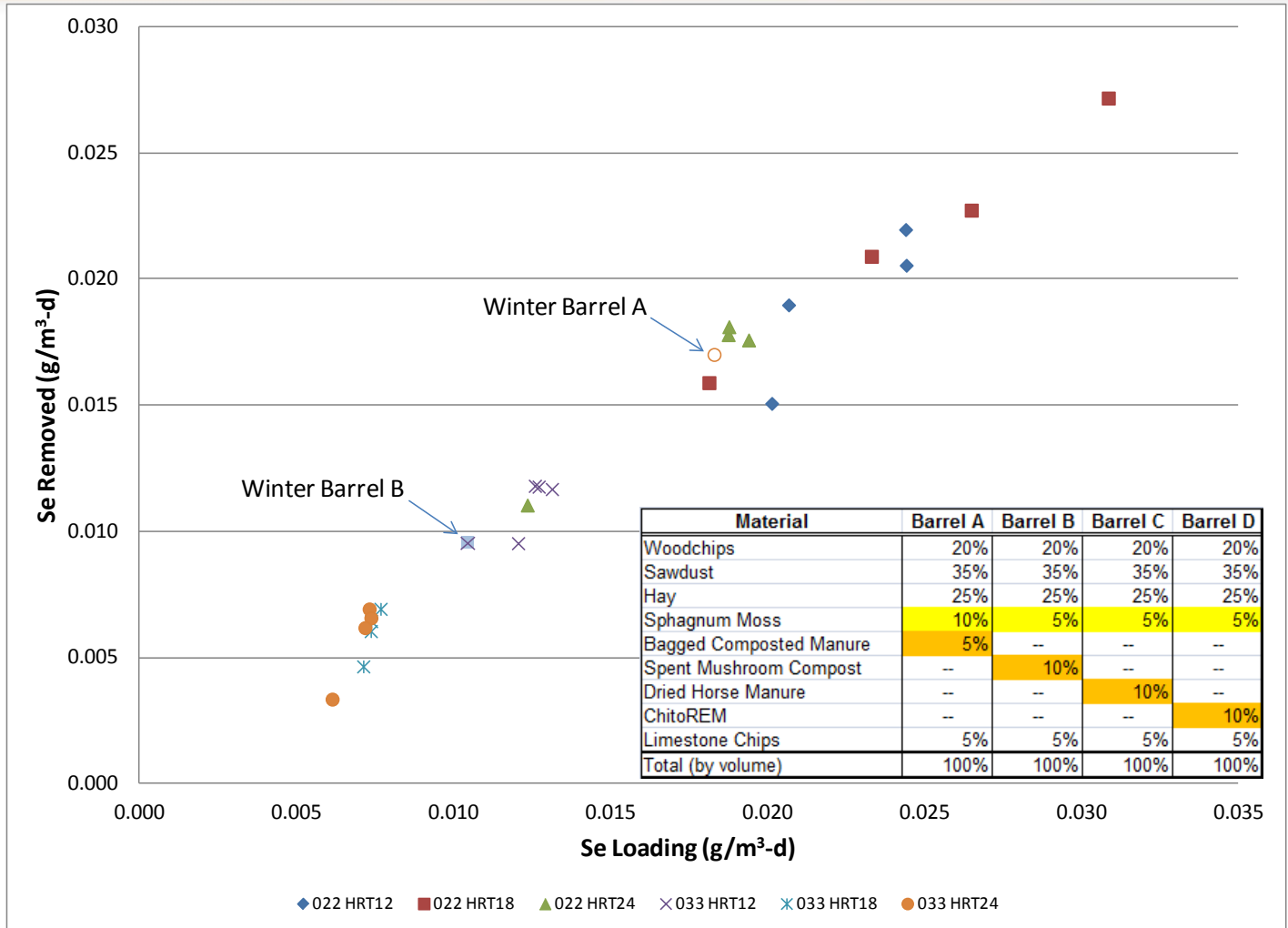


**Outlet 033 - Barrel D**  
Avg. Se Mass ( $\mu\text{g}/\text{L}$ )





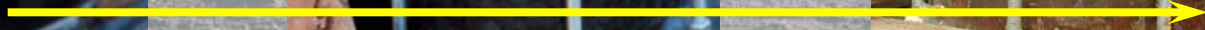
# Se Mass Removed vs Volumetric Loading



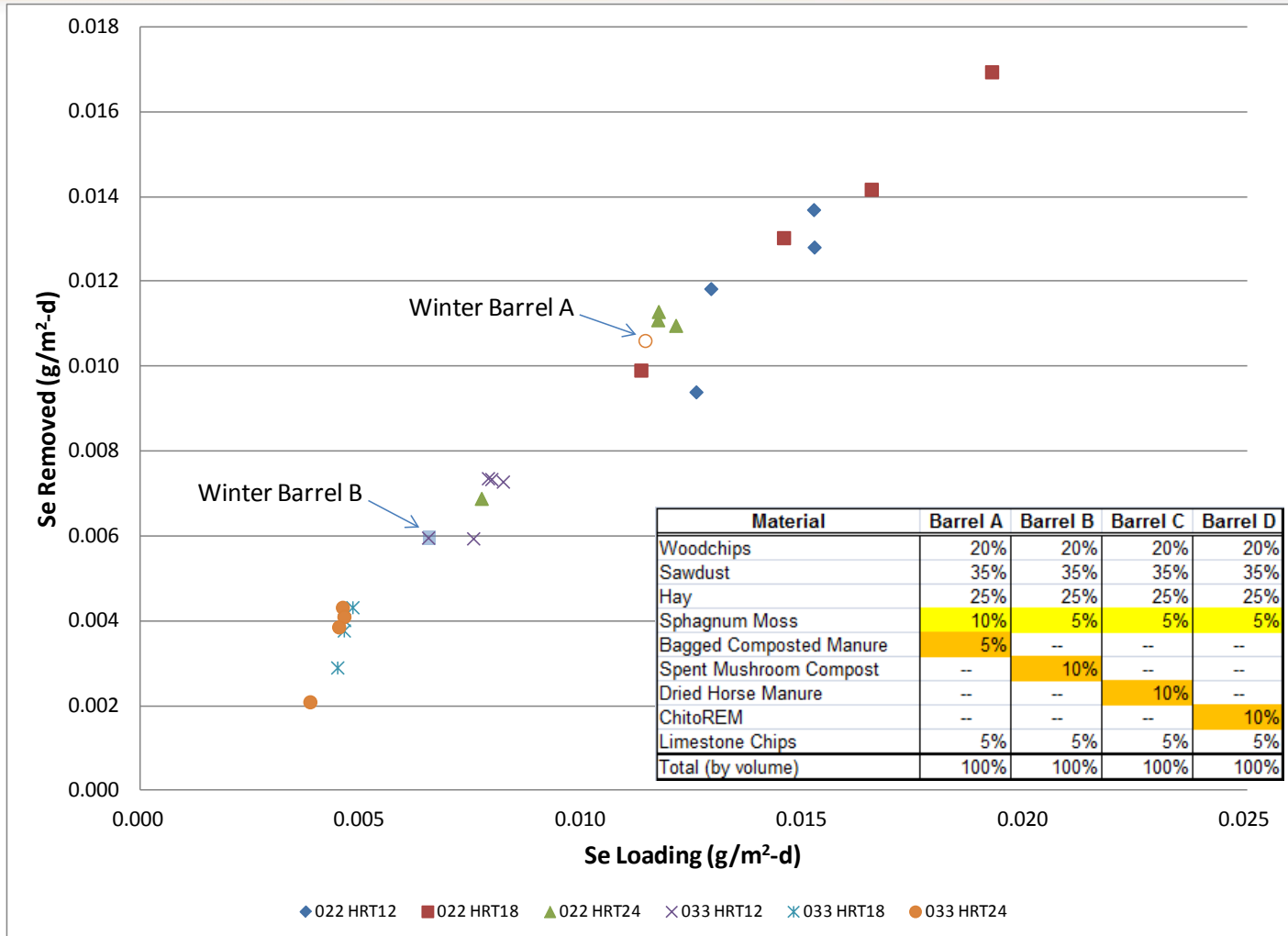
# Reaction Fronts in BCR Substrates Over Time



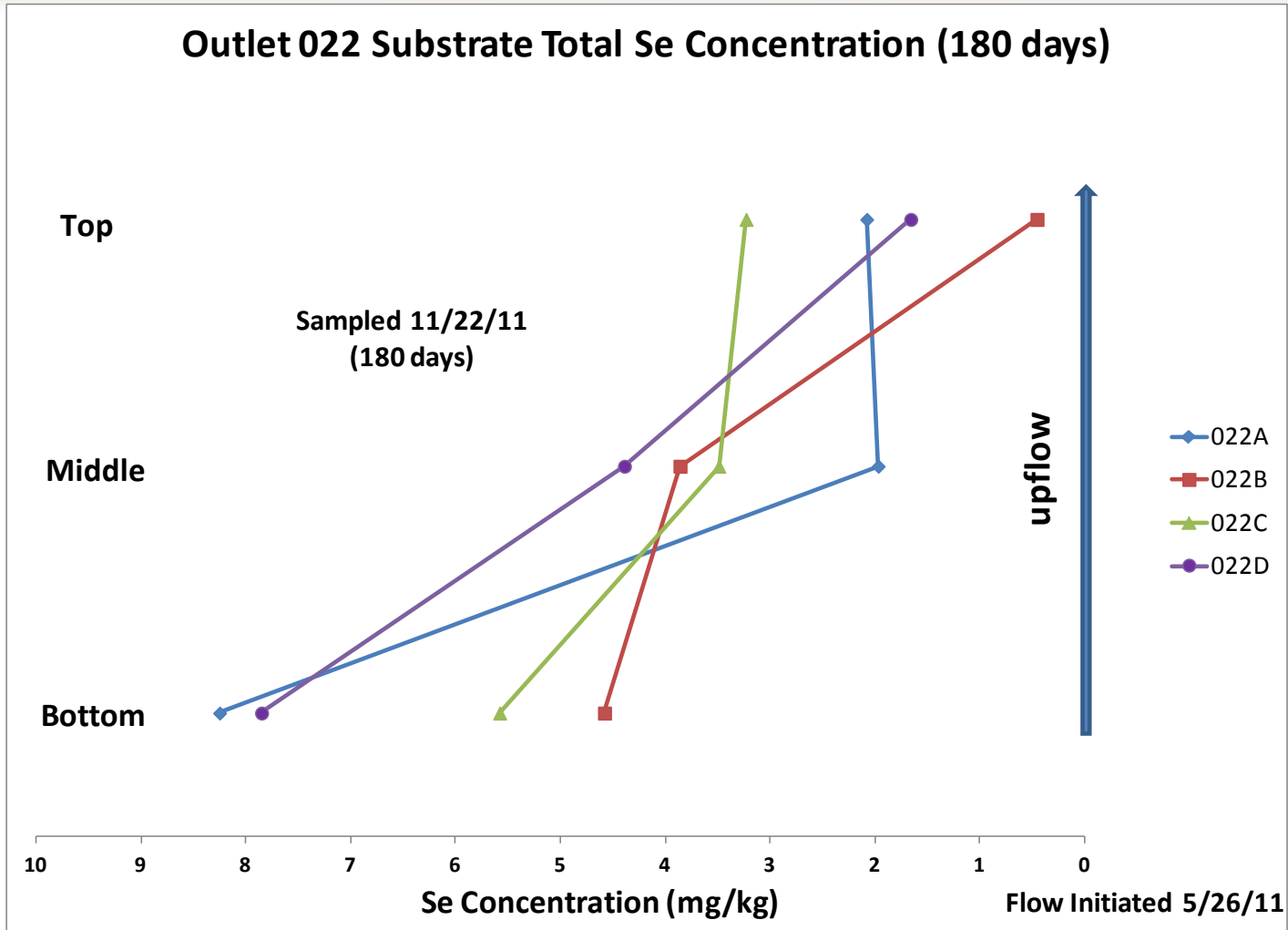
Time



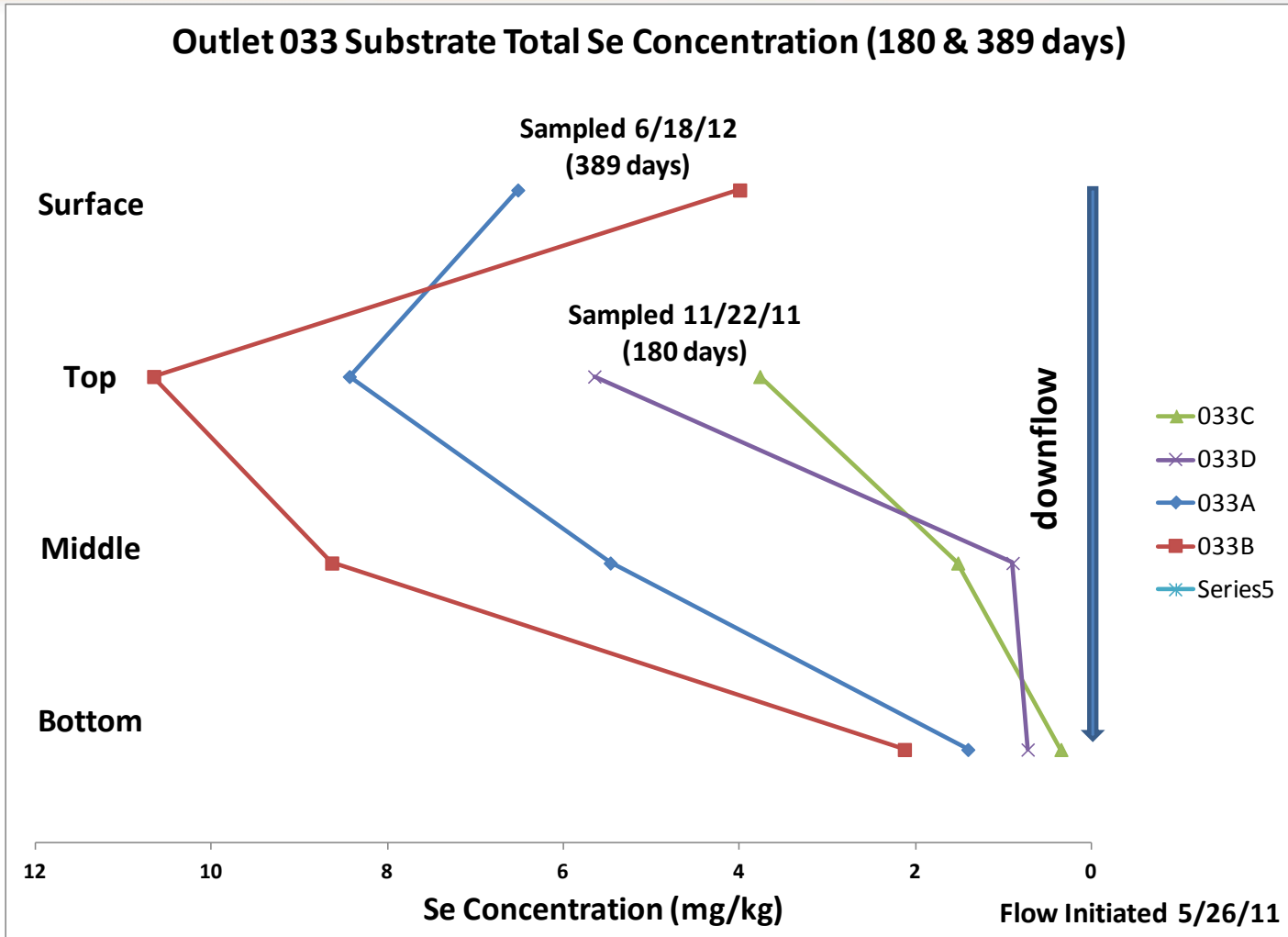
# Se Mass Removed vs Surface Aerial Loading



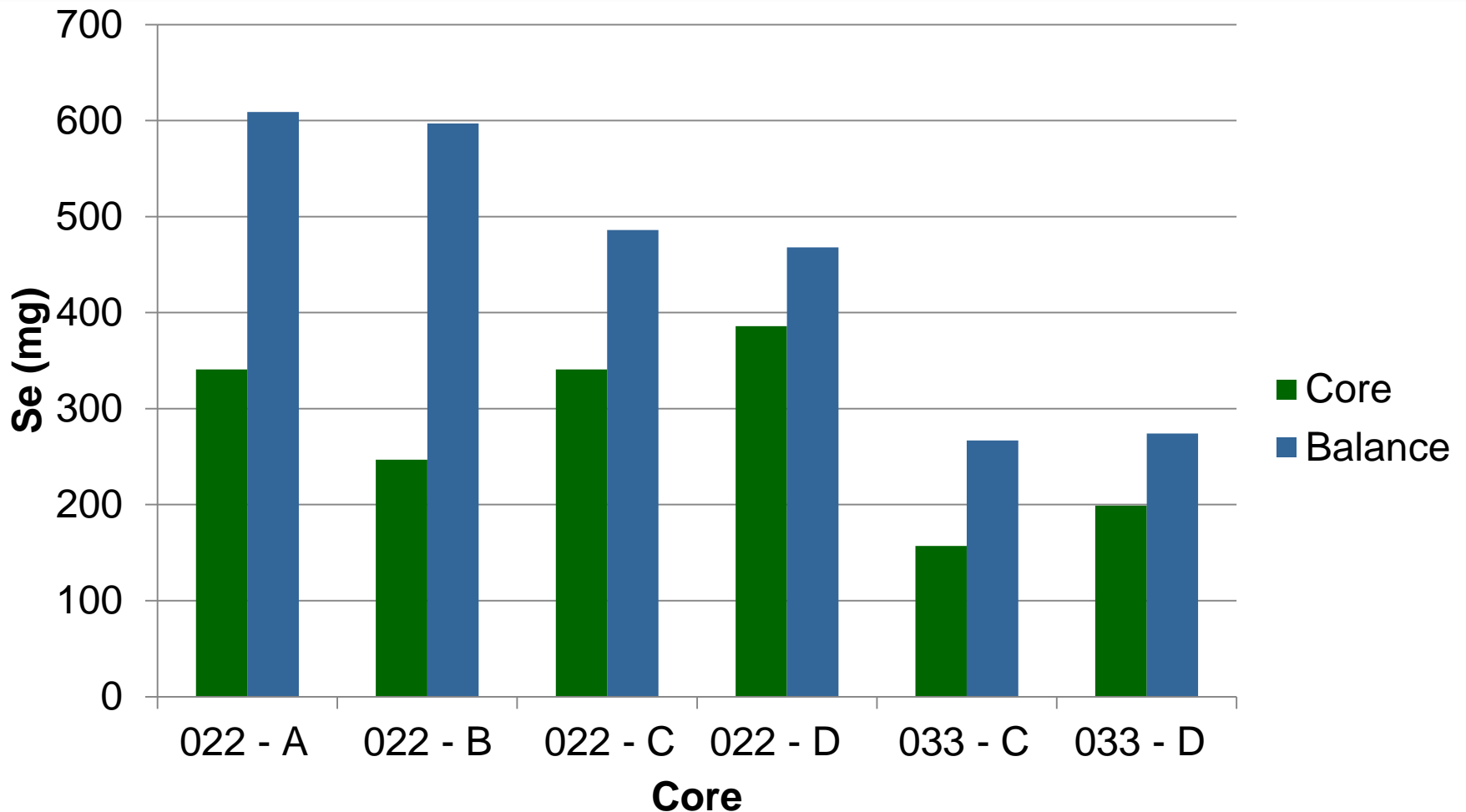
# Barrel Selenium Profile



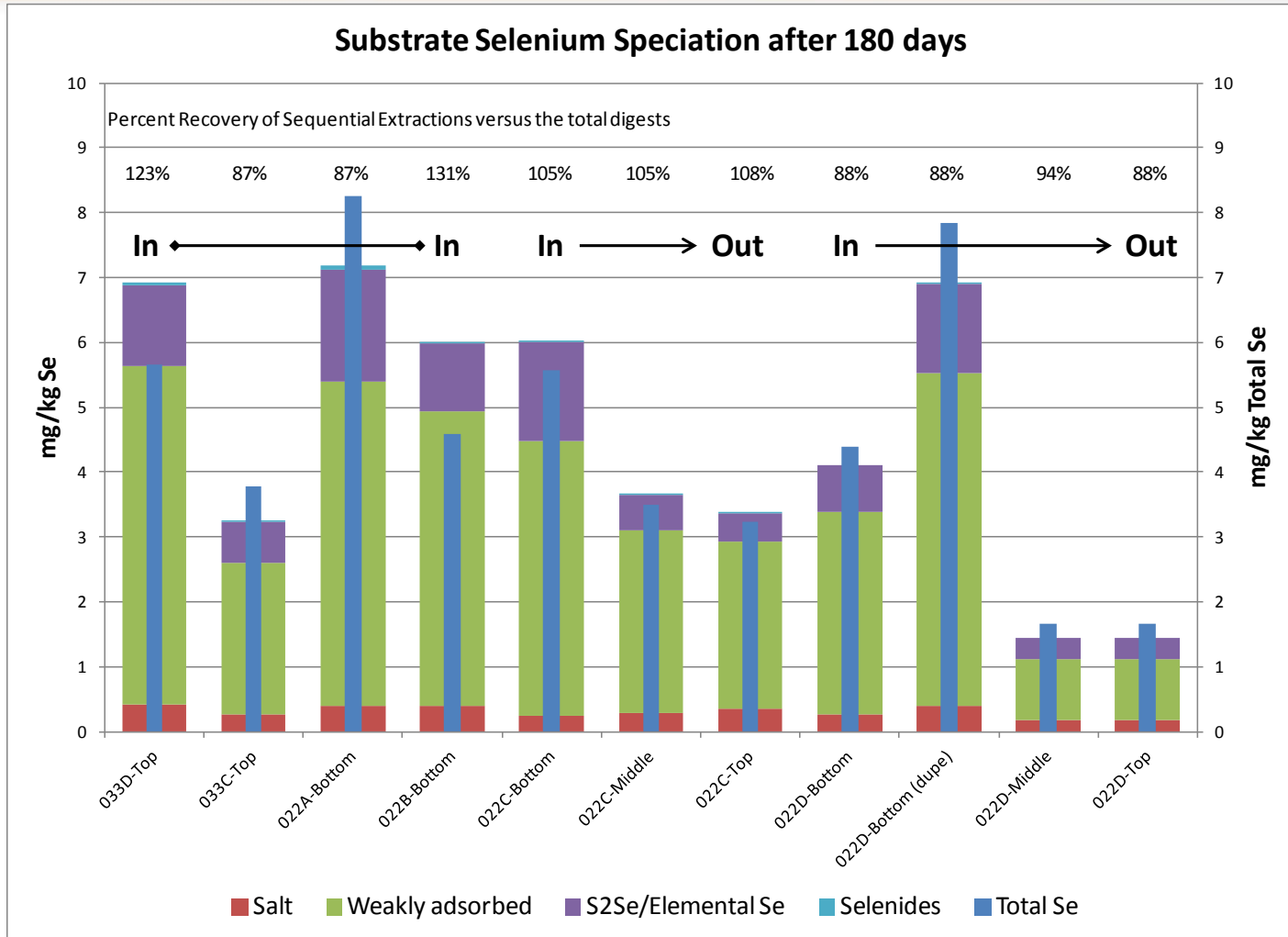
# Barrel Selenium Profile



# Selenium Mass Balance vs Sediment Core: Gap Suggests Volatilization Loss

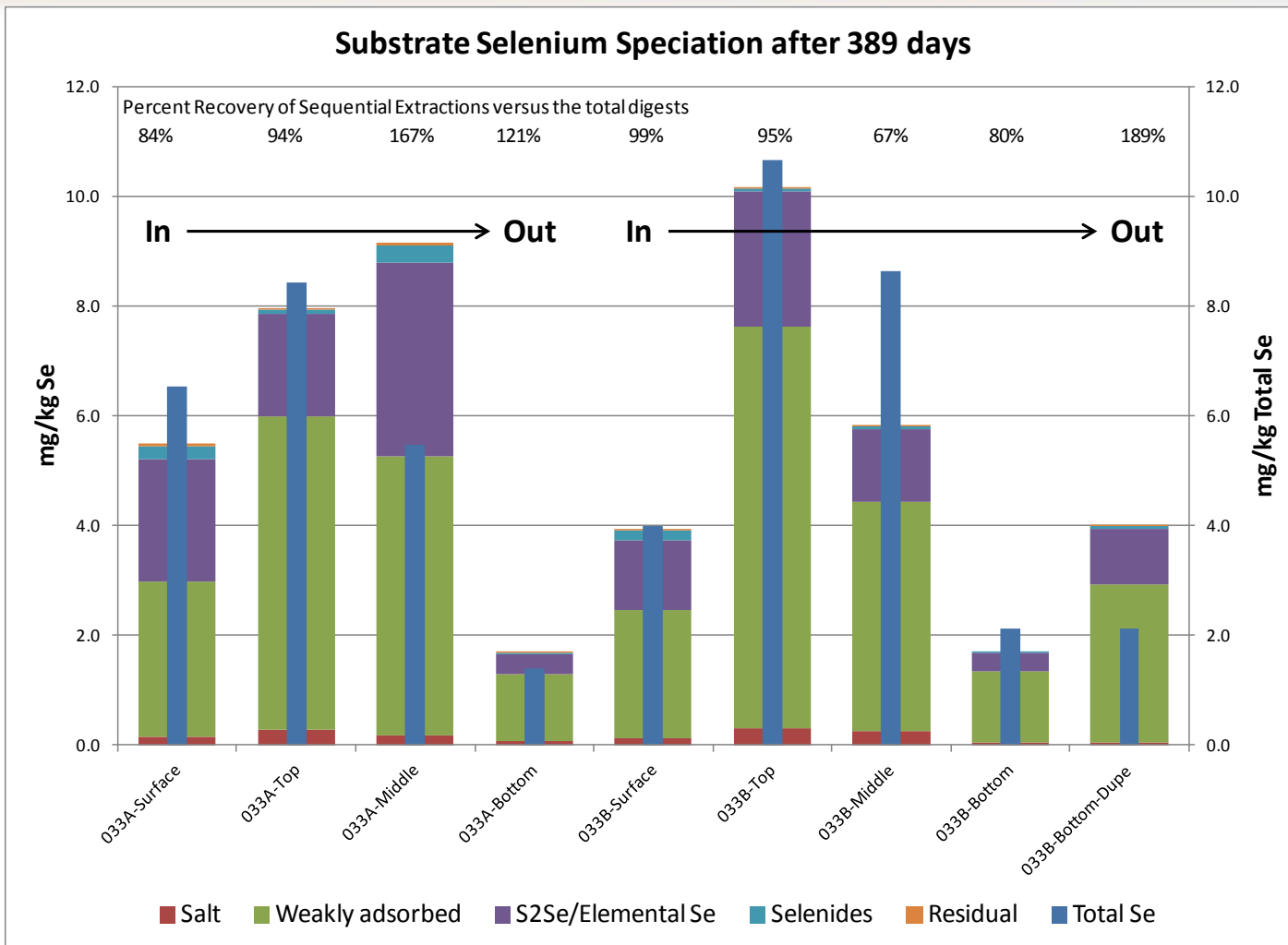


# Vertical Distribution and Speciation of Selenium: Reduction, Sorption, Volatilization



Source:  
CH2MHILL (2012)

# Vertical Distribution and Speciation of Selenium: Reduction, Sorption, Volatilization



Source:  
CH2MHILL (2012)





# Model Rate Constant Calibration

- Empirical data from barrel studies used to calibrate treatment rates in the development of a sizing model for a full-scale treatment system
- General terms of mass conservation for a plug flow system are used while maintaining the first order model as the basis
- Empirical data was modeled using first order area-based treatment wetland model of Kadlec & Knight (1996) & Kadlec & Wallace (2009)
- Model estimates the potential BCR effluent concentration for a given flow rate and concentration for a given BCR area, as adjusted for temperature

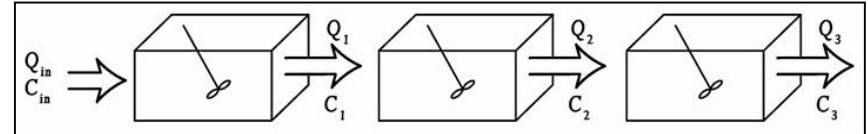
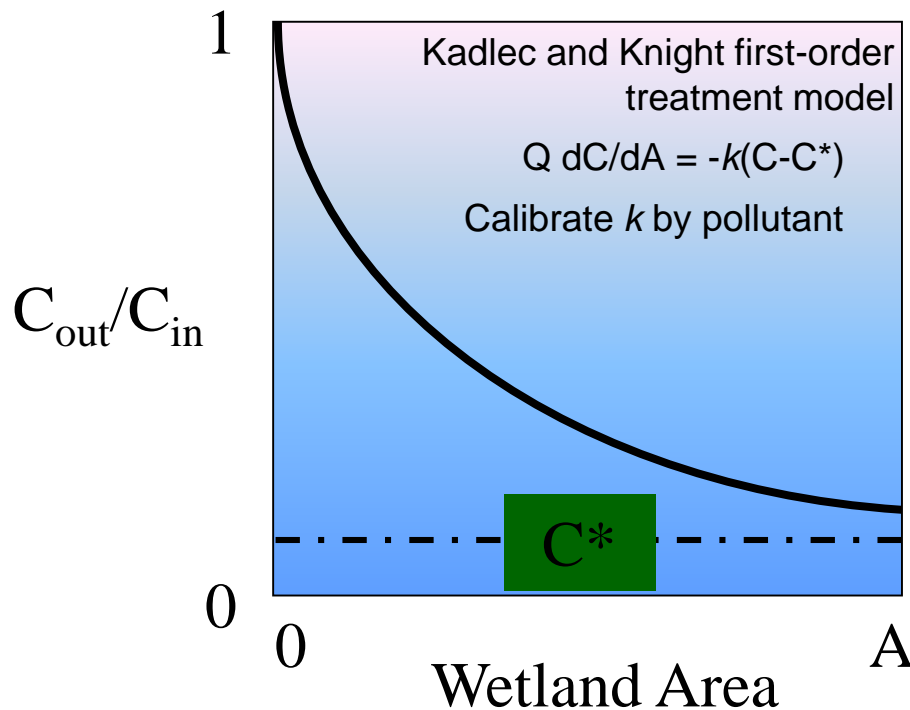
$$J = k(C_i - C_e)$$

where:

- $J$  = zero-order contaminant removal rate [ $\text{g}/\text{m}^2/\text{yr}$ ]
- $k$  = first-order, area-based rate constant ( $\text{m}/\text{yr}$ )
- $C_i$  = influent concentration ( $\text{g}/\text{m}^3$ )
- $C_e$  = effluent concentration ( $\text{g}/\text{m}^3$ )

# First-order Area-based Treatment Model

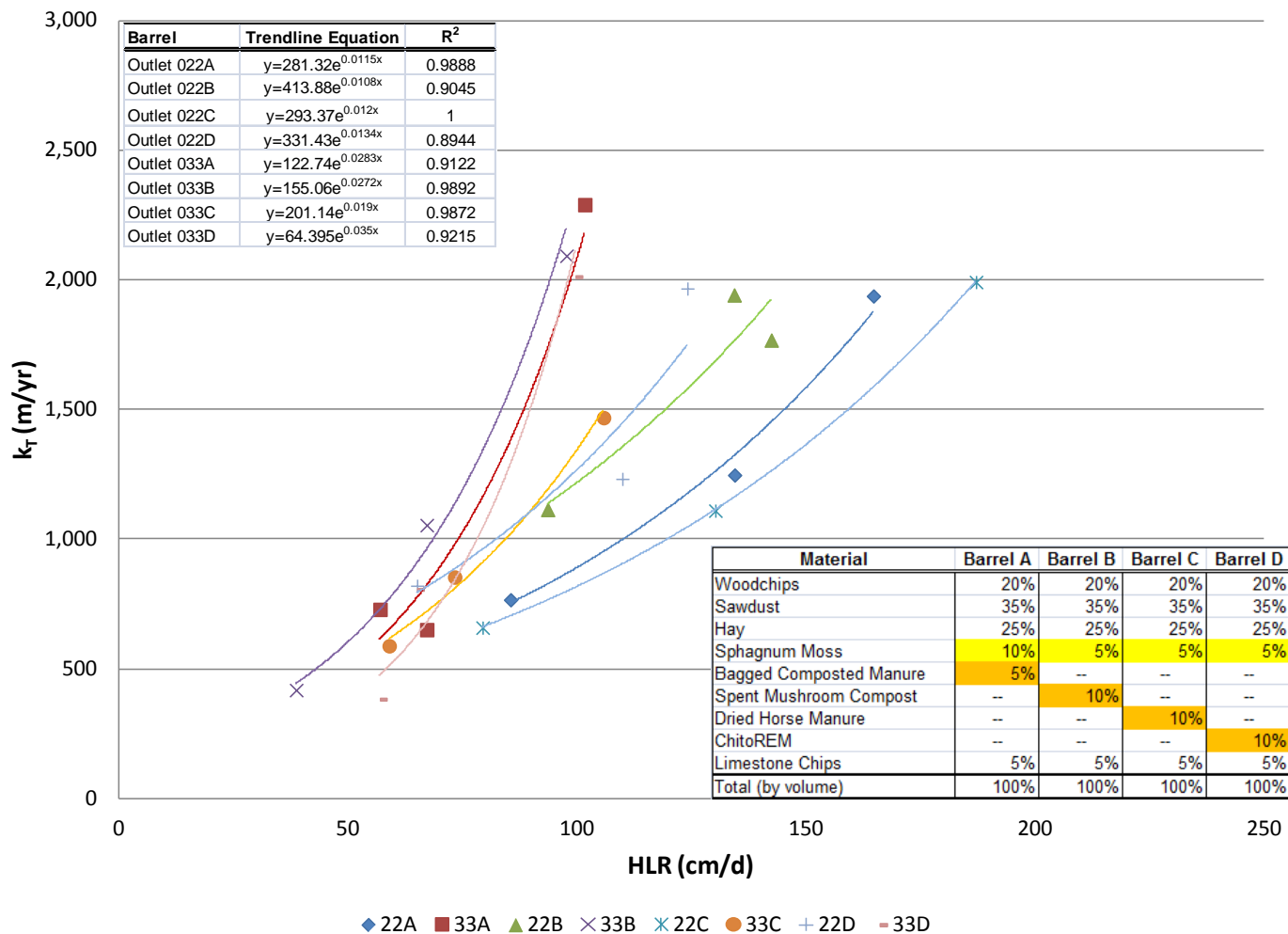
First-order treatment model was expanded to the P-k-C\* model, which is solved by relating: 1) hydraulic loading, 2) removal rate, 3) concentration terms, and 4) hydraulic mixing



$$\left( \frac{C - C^*}{C_i - C^*} \right) = \frac{1}{(1 + k/Pq)^P}$$

Green arrows point from the  $C_i$  term in the numerator to the  $C$  term in the numerator, from the  $P$  term in the denominator to the  $P$  term in the denominator, and from the  $q$  term in the denominator to the  $Q$  term in the denominator.

# Rate Constant vs Hydraulic Loading Rate



# First-Order, Area-Based Tanks-in Series Rate Constant Modeling Results

- Passive Treatment Systems rarely operate as plug-flow reactors
  - Heterogeneous substrate, preferential flow
- Passive Treatment Systems are better characterized hydraulically as a series of well-mixed tanks (3 – 6 in series for VFW)
- Barrel-study data was configured in a three tank series using the P-k-C\* model to assess the variable hydraulic characteristics on overall performance

**First-Order, Area-Based Tanks-in Series Rate Constant Modeling Results**

Barrel	1st-order, Area Based, Rate Constants (m/y)			
	Outlet 022 <sup>a</sup>	Outlet 033 <sup>a</sup>	Outlet 033 <sup>b</sup>	Outlet 033 <sup>c</sup>
A	1,379	1,498	675	675
B	1,744	1,671	1,177	1,177
C	1,246	1,180	—	—
D	1,547	1,115	—	—

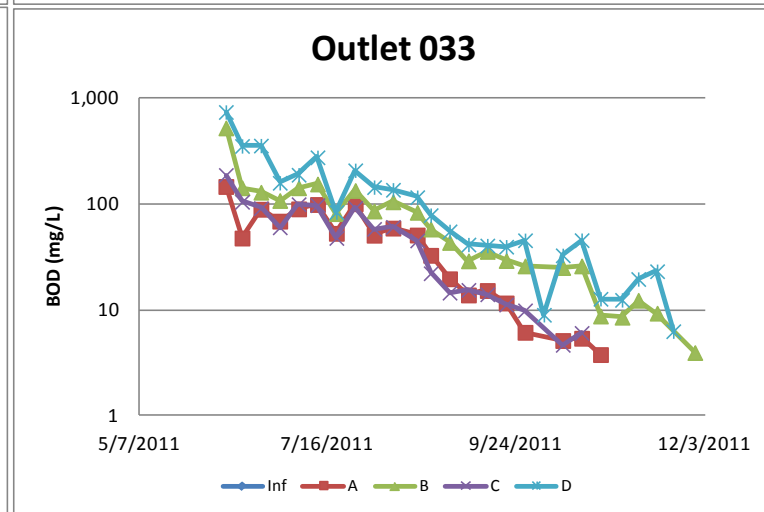
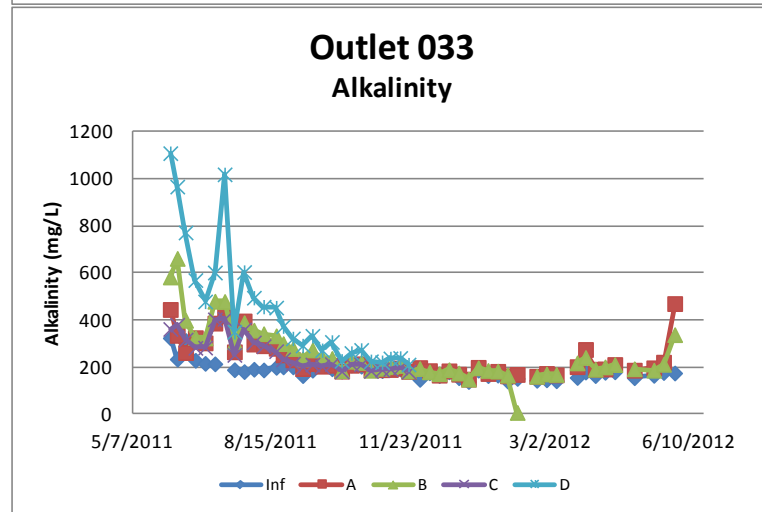
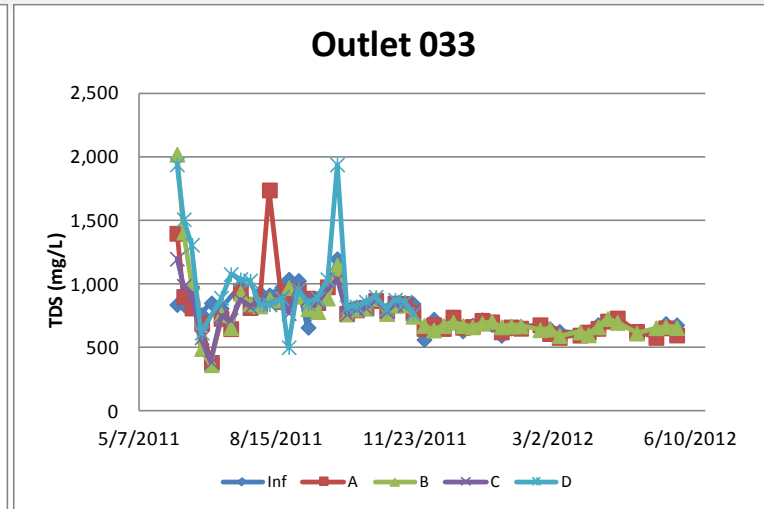
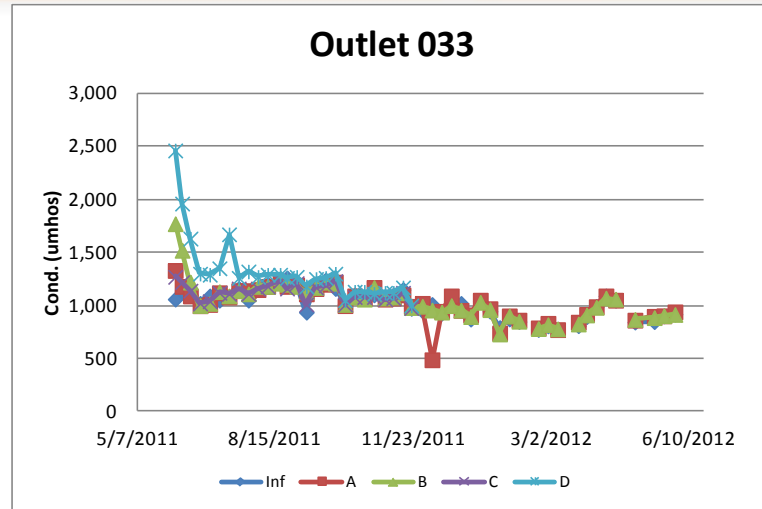
*Note:* Target effluent concentration was set at 4 µg/L.

<sup>a</sup> Influent temperatures monitored during the summer and fall operations used in model ( $\theta$  set at 1).

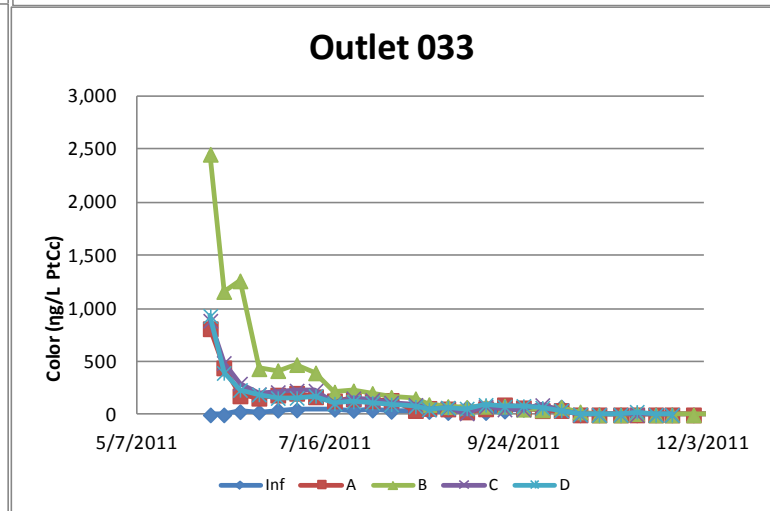
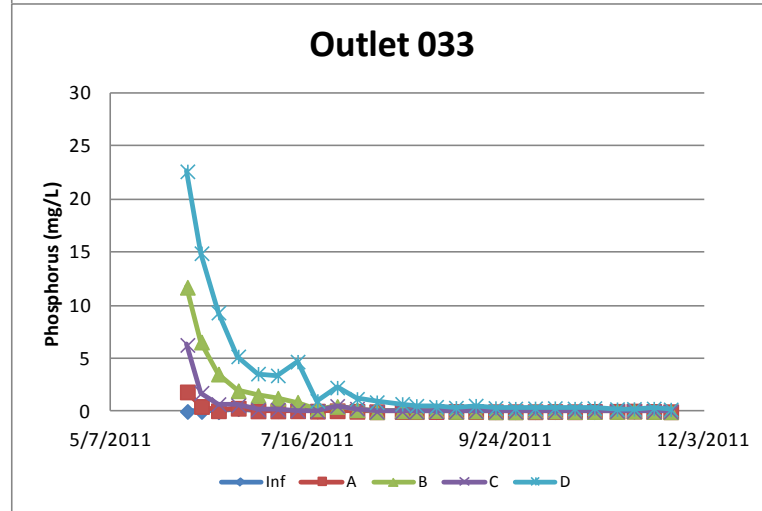
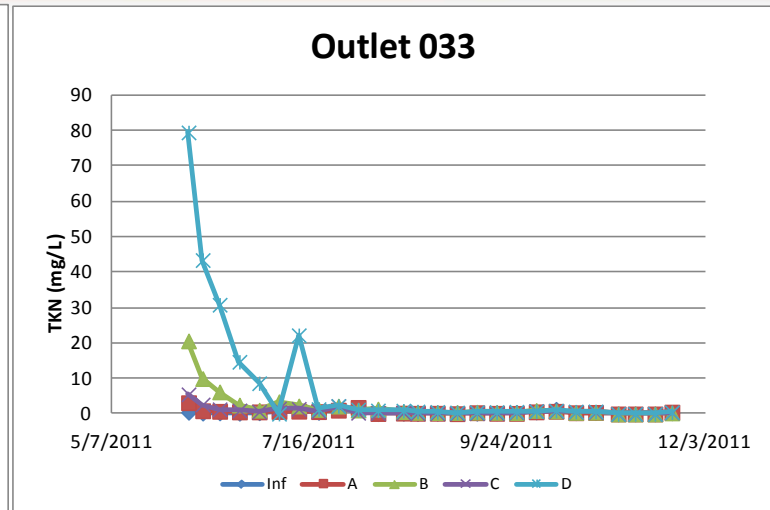
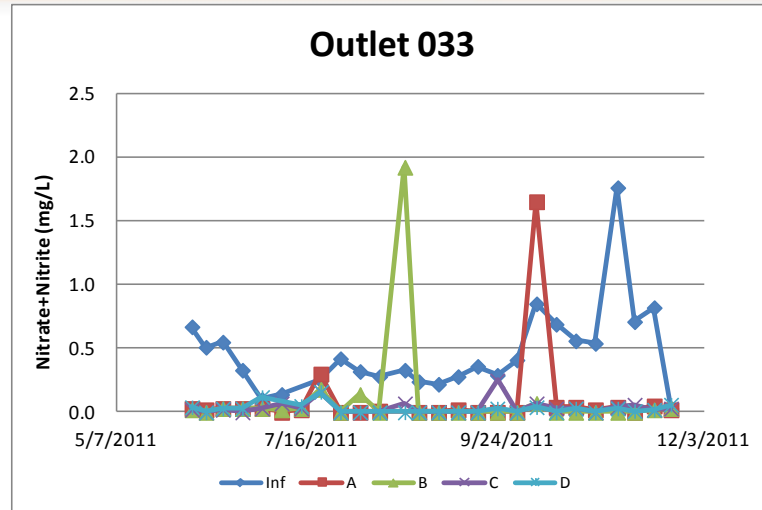
<sup>b</sup> Influent temperatures monitored during the winter operations used in model (10.5°C) ( $\theta$  determined to be 0.95).

<sup>c</sup> Influent temperature set at 9°C in model ( $\theta$  determined to be 0.96).

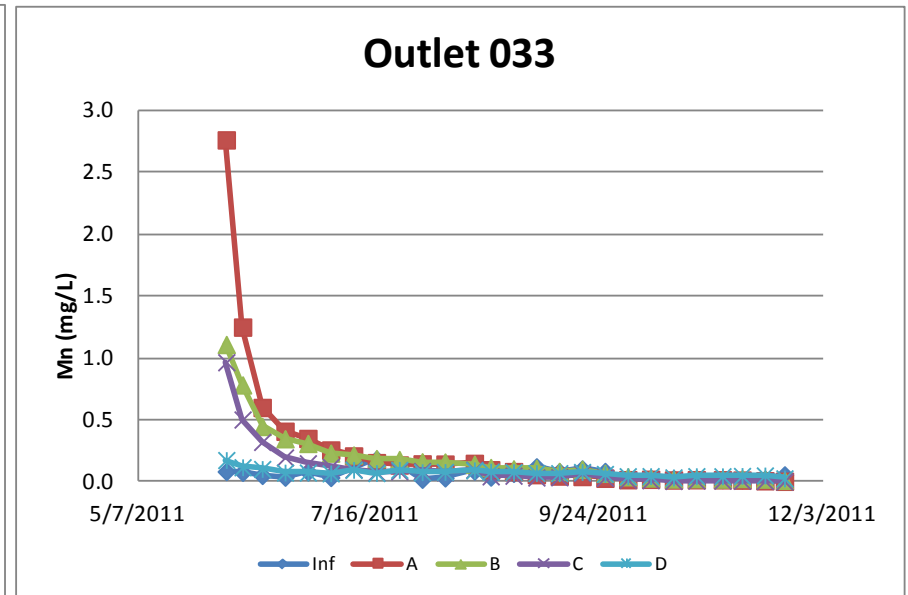
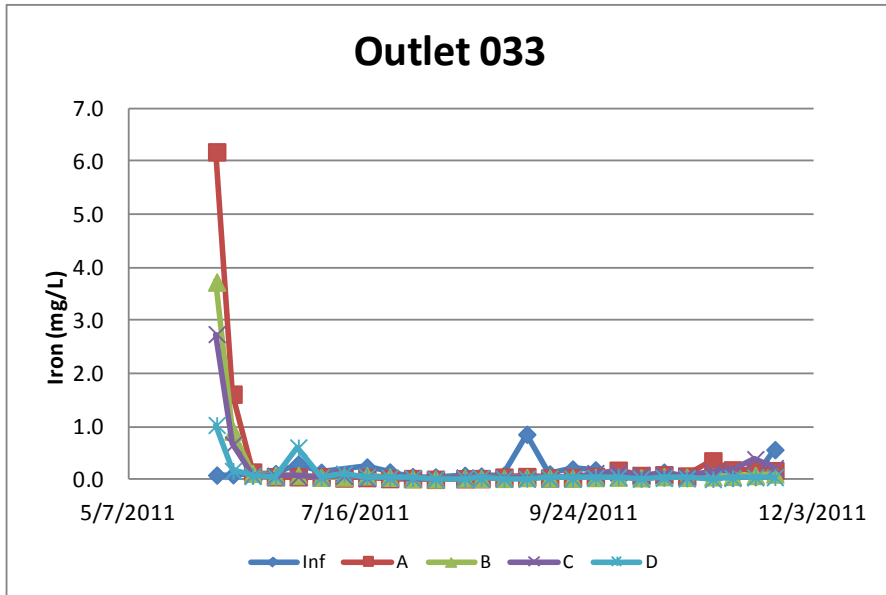
# Secondary Parameters (Salts & Labile C): Conductivity, TDS, Alkalinity, BOD



# Secondary Parameters: Nutrients and Color



# Secondary Parameters: Metals





# Conclusions

- Selenium was effectively reduced by ~90% on average in all barrels.
- Selenium removal was effective at all HRTs tested (12 to 24 hours).
- Cold temperatures (below freezing) did not affect selenium removal rates significantly, and effluent selenium concentrations remained below discharge limits following startup.
- The organic substrates tested could generate secondary byproducts:
  - short-term increase in conductivity from the flushing of excess salts from the substrate during startup
  - elevated oxygen demand from excess labile carbon in the effluent
  - suspended solids (mainly fine particulate organic matter)
  - excess nutrients in the substrate mixture
  - release of regulated metals such as iron and manganese, which might also occur in small amounts in the substrate mixture.





# Conclusions

- Se removal occurs at the influent water-substrate interface
  - consistent with first-order processes
  - extent of selenium distribution within the substrate is dependent on the selenium load.
- Substrate Se speciation indicates:
  - dominant mechanism is reduction of selenate to reduced forms of selenium that are weakly adsorbed to the substrate during early stages of selenium removal
  - approximately >50% weakly adsorbed Se attributable to selenite
  - Highly immobile elemental selenium or selenosulfide account for about a quarter of the total selenium retained
  - very little metal selenide was found.
- Both substrate coring and water-balance mass-balance evaluation methods show selenium retention
  - higher water balance method suggests a higher overall selenium retention than that observed in the substrate cores, suggesting loss to volatilization.
- Barrel media type B appeared to balance high treatment performance with moderate production of byproducts.
- Selenium treatment BCRs have been shown to treat mine site discharges effectively in order to achieve selenium compliance limits.



# Acknowledgements

- Thanks to Alpha Natural Resources
- Thanks to supporting engineering and science staff at CH2M HILL



# Questions